

ROBUST SUMMARY FOR AMINOALKYLNITRILE CATEGORY 7: 35

Summarv

For purposes of this HPV submission, the aminoalkylnitrile category is composed of two chemicals with two functional groups, an amino group and a nitrile group, both of which are bonded to the same carbon atom. This carbon atom also bears a methyl group and another alkyl group. This category is composed of discrete materials that change by an incremental increase in carbon number in the alkyl moiety. The aminoalkylnitriles included in this HPV category are 2-amino-2-methylpropanenitrile and 2-amino-2-methylbutanenitrile. The next higher homologue, 2-amino-2,3-dimethylbutanenitrile has been the subject of a separate HPV submission. Because of the close similarity in the structure and properties of this homologue, it will be considered as a supporting analog, and data on this analog are used to supplement data for the aminoalkylnitrile category.

For purposes of this HPV document, the aminoalkylnitrile chemicals can be represented by the general structural formula:

$$\begin{array}{c} \mathsf{CH_3} \\ \mathsf{R-C-CN} \\ \mathsf{NH_2} \end{array}$$

Information regarding these chemicals is presented in the table below.

<u>Chemical Name</u>	CAS Registry Number	<u>R =</u>
Propanenitrile, 2-amino-2-methyl-	19355-69-2	CH ₃ - (Category Member)
Butanenitrile, 2-amino-2-methyl-	4475-95-0	CH ₃ CH ₂ - (Category Member)
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The members of this category are produced solely by DuPont, as company-limited intermediates for the synthesis of the corresponding azonitriles, 2,2'azobis-(2-isobutyronitrile) (AIBN) (CAS # 78-67-1) and 2,2'azobis-(2-methylbutyronitrile) (AMBN) (CAS #13472-08-7). An HPV submission was made to EPA for AMBN, and in this submission AIBN was proposed as an analog to provide data to support AMBN. Because of the similar molecular structures, comparable effects data, and expected similar metabolic pathway, EPA agreed that AIBN is an acceptable analog for AMBN. We believe that similar considerations also justify treating 2-amino-2-methylpropanenitrile and 2-amino-2-methylbutanenitrile as members of an HPV

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category, and justify using data for 2-amino-2,3-dimethylbutanenitrile to support this aminoalkylnitrile category.

The scientific literature was searched and summarized. Data were identified for the two materials in the category and the analogous substance (Table 1). Each study on category materials was evaluated for adequacy. EPA has already evaluated the HPV submission for the supporting analog. Robust summaries were developed for each study addressing specific SIDS endpoints. Summaries were also developed for studies that were either considered not adequate but provided information of relevance for hazard identification and evaluation, or covered non-SIDS endpoints (Appendices A-C).

Table 1: Matrix of Available and Adequate Data for Aminoalkvlnitrile Category

	Propanenitrile, 2- amino-2-methyl- (Category Member)	Butanenitrile, 2- amino-2-methyl- (Category Member)	No. 2011 No. West Projection of the Control of the
R =	CH ₃ -	CH₃CH₂-	
PHYSICAL/CHEMICAL CHARACTERIST		· · · · · · · · · · · · · · · · · · ·	·
Melting Point	√/-	√/-	
Boiling Point	√/-	√/	
Vapor Pressure	√/_	√/	
Partition Coefficient	√		
Water Solubility	√/_	√/	
ENVIRONMENTAL FATE			
Photodegradation	1 1	√	TOTAL METERS AND
Stability in Water	\/_	\	
Transport (Fugacity)		7	
Biodegradation	V/-	√/_	
ECOTOXICITY			
Acute Toxicity to Fish (96-hour LC ₅₀)	\ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \	V	
Acute Toxicity to Invertebrates (48-hour	V	-	
EC ₅₀)	<u> </u>		
Acute Toxicity to Aquatic Plants			
MAMMALIAN TOXICITY			
Acute Toxicity	√	T	
Repeated Dose Toxicity	\/-		
Developmental Toxicity			
Reproductive Toxicity	_		
Genetic Toxicity Gene Mutations	<u> </u>		
Genetic Toxicity Chromosomal			
Aberrations			
√ = Data are available and considered adequate. √ = Data are available, but considered inadequate.	nate.	····	
- = No data available. N/A = Not Applicable.			

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All three nitriles have roughly equivalent physical chemical properties (Table 2). Molecular weights range from 84.12 to 112.17. They are all liquids at room temperature, with melting points ranging from -4.7 to 7.7°C, and all three decompose with heat. Since the estimated melting point values was above 0°C for 2-amino-2-methylbutanenitrile was above 0°C and the value for 2-amino-2-methylpropanenitrile was only slightly below 0°C, measuring melting points of these 2 chemicals following OECD Guideline 102 is recommended. In addition, the test samples will be observed for evidence of decomposition during the melting point measurement. Measured vapor pressure values are 30 mm Hg at 66°C, 14 mm Hg at 68°C, and 23.42 mm Hg at and 2-amino-2.3dimethylbutanenitrile, respectively. Estimated vapor pressures are also included at the standard temperature of 25°C, where measured data at this temperature were not available. Although measured values for vapor pressure are supplied, no additional data regarding the testing of this endpoint were available. Therefore, vapor pressure studies for 2-amino-2-methylbutanenitrile and 2-amino-2-methylpropanenitrile following OECD Guideline 104 are recommended, contingent upon the technical feasibility of obtaining test material of sufficient purity and stability for running these studies. Estimated vapor pressures were used when needed in modeling environmental fate data. Although no density was reported for 2-amino-2.3dimethylbutanenitrile, the density for 2-amino-2-methylpropanenitrile and 2-amino-2methylbutanenitrile are similar, with values of 0.9 and 0.886, respectively. Partition coefficients are similar with estimated values of -0.04, -0.25, and 0.87 for 2-amino-2-methylpropanenitrile, 2amino-2-methylbutanenitrile, and 2-amino-2,3-dimethylbutanenitrile, respectively. All three aminoalkylnitriles show appreciable water solubility with values greater than or equal to 27 g/L. Additional information regarding the water solubility of 2-amino-2-methylbutanenitrile and 2amino-2-methylpropanenitrile will be determined in conjunction with testing of stability in water. The available data show similarity between the three nitriles for physical and chemical characteristics, thus supporting the category approach. Additional testing for physical/chemical characteristics recommended for 2-amino-2-methylpropanenitrile and 2-amino-2-methylbutanenitrile include melting point (OECD Guideline 102), boiling point (decomposition data), vapor pressure (OECD Guideline 104, contingent upon technical

feasibility of running the studies), and water solubility (data to be collected in conjunction with testing of stability in water).

(log Kow)

Propanenitrile, 2-Butanenitrile, 2amino-2-methylamino-2-methyl-Brown liquid with Yellow liquid with **Physical Appearance** an ammonia-like an ammonia-like odor odor Molecular 84.12 98.15 Weight Water 27.0 g/L **Solubility** -4.7°C 7.1°C Melting Point **Boiling Point** Decomposes Decomposes 14 mm Hg (a), 68°C Vapor 30 mm Hg @ 66°C Pressure (measured) (measured) 4 mm Hg @ 20°C (measured) 2.84 mm Hg @ 1.03 mm Hg @ 25°C (estimated) 25°C (estimated) 0.9 @ 25°C Density/ 0.886 **Specific** Gravity **Partition** -0.04 (estimated) 0.45 (estimated) Coefficient

Table 2: Physical and Chemical Characteristics

Members of the aminoalkyhritrile category have similar environmental fate behavior (Table 3). At acidic to neutral environmental pH, all three aminoalkyhritriles may be ionized due to the presence of the amino group, then subject to cation exchange reactions. Although somewhat volatile, with vapor pressures above 0.1 mm Hg (Table 2), they have Henry's Law constants less than so there will be a tendency to rain out of the atmosphere and not to volatilize from surface waters. Based on the atmospheric oxidation models, the two substances in the category have estimated half-lives of greater than 10 days, due to hydroxyl radical oxidation. The category analog is subject to the same oxidation mechanism, but with a higher hydrogen:carbon ratio is oxidized more rapidly, with an estimated half-life of 1.85 days. All three aminoalkylnitriles show appreciable water solubility with values greater than or equal to 27 g/L (Table 2). They are likely to be unstable in water, however, because they show a tendency to disproportionate to the corresponding ketone, cyanide, and ammonium when dissolved in water in the absence of excess ammonia (Kirk-Othmer, 1978). Measured stability in water testing for 2-amino-2-methyl-propanenitrile and 2-amino-2-methylbutanenitrile following OECD Guideline 111 is recommended. Biodegradation is estimated to be fast for

2-amino-2-methylpropanenitrile and 2-amino-2-methylbutanenitrile. Because of the presence of a dimethyl group, the model estimates that the supporting analog is not as readily biodegradable. Since no measured biodegradation data is available, biodegradation testing following OECD Guideline 301B or 302B is recommended for 2-amino-2-methylpropanenitrile and 2-amino-2methylbutanenitrile, dependent upon the results of the stability in water testing. Substances degraded by water alone would also be expected to degrade at the same rate or faster in biologically active soil in the presence of water. Therefore, biodegradation testing is recommend only if the results of the stability in water testing at pH 7.0 demonstrate half-lives of the parent chemicals are greater than 38 days. The category shows little tendency to bioaccumulate based on low estimated BCF values. Consistent with behavior described above, and assuming equal emissions to air, water, and soil, any residual of the aminoalkylnitrile category is expected to be distributed primarily in water and soil, based on the Mackay Level III fugacity model. Therefore, with regard to expected environmental distribution, the aminoalkylnitriles behave in a similar manner, justifying their classification as a category. Additional testing for environmental fate recommended for 2-amino-2-methylpropanenitrile and 2-amino-2methylbutanenitrile include stability in water (OECD Guideline 111) and biodegradation (dependent upon results of stability in water).

Propanenitrile, 2-Butanenitrile, 2amino-2-methylamino-2-methyl-Bioaccumulation a log BCF = 0.5log BCF = 0.5Biodegradation* Readily Readily degradable degradable Fugacity* Air 0.1% Air 1% Water 45.9% Water 44.8% Soil 53.9% Soil 55% Sediment 0.09% Sediment 0.089% Modeled data.

Table 3: Environmental Fate

The nitriles are moderately to highly toxic to aquatic life (Table 4). 2-Amino-2-methyl propane&rile, 2-amino-2-methylbutanenitrile, and 2-amino-2,3-dimethylbutanenitrile are highly toxic to fish with a 96-hour LC_{50} of 0.71 to 0.75 and 2-amino-2,3-dimethylbutanenitrile are moderately toxic to Daphnia with 48-hour EC_{50} 's of 6.9 and 7.1 mg/L, respectively. 2-Amino-2,3-dimethylbutanenitrile is highly toxic to algae with a 96-hour EC_{50} of 0.36 mg/L. The three chemicals appear to have somewhat similar toxicity to

a 96-hour EC₅₀ of 0.36 mg/L. The three chemicals appear to have somewhat similar toxicity to the individual species. Some differences exist, with algae appearing to be more sensitive than fish or invertebrates. The available data are similar for all three nitriles, supporting the category approach for ecotoxicity. Since the database indicates that there is strong agreement in aquatic toxicity across the category and analog chemicals, and data exists for each study type, no additional ecotoxicity testing is recommended.

Table 4: Ecotoxicity

	Propanenitrile, 2- amino-2-methyl-	Butanenitrile, 2- amino-2-methyl-	
Toxicity to Fish (96-hour	0.71 mg/L (N)	0.71 mg/L (N)	
LC ₅₀ value)	468.3 mg/L (E)	744.5 mg/L (E)	and the second s
Toxicity to Invertebrates	7.1 mg/L (N)		
(48-hour EC ₅₀ value)	26.6 mg/L (E)	41.1 mg/L (E)	
Toxicity to			
Algae (96-hour EC ₅₀ value)	24.8 mg/L (E) n nominal test concentr	35.7 mg/L (E)	

E =estimate value; log Kow values used in the ECOSAR model are listed in Table 2.

Acute toxicity data indicate that the three chemicals exhibit similar acute toxicity (Table 5). 2-Amino-2-methylpropanenitrile is very toxic to mammals with an oral LD₅₀ in rats of 1 O-30 mg/kg; while 2-amino-2-methylbutanenitrile and 2-amino-2,3-dimethylbutanenitrile are toxic with oral LD₅₀s of 74 and 83 mg/kg, respectively. All three chemicals are toxic via the inhalation route with a 1-, 2-, and/or 4-hour ALC (approximate lethal concentration) or LC₅₀ ranging from 7 1 - 111 ppm. Dermally, 2-amino-2-methylpropanenitrile and 2-amino-2,3dimethylbutanenitrile are very toxic with an ALD (approximate lethal dose) and LD50 in rabbits of 30-100 and 23 mg/kg, respectively. The test substances produced slight to mild skin irritation. 2-Amino-2-methylpropanenitrile and 2-amino-2,3-dimethylbutanenitrile produced mortality when tested in rabbit eyes. 2-Amino-2-methylbutanenitrile did not cause death of rabbits, but was a severe eye irritant. 2-Amino-2-methylbutanenitrile was not a skin sensitizer when tested in guinea pigs. No data regarding the acute dermal toxicity of 2-amino-2-methylbutanenitrile, or dermal sensitization potential of 2-amino-2-methylpropanenitrile and 2-amino-2,3dimethylbutanenitrile were available. The available acute toxicity data are similar for the three nitriles, thus supporting the category approach for acute toxicity. All required SIDS acute toxicity data points are complete for the category, and no further acute mammalian testing is recommended.

Propanenitrile, 2-Butanenitrile, 2amino-2-methylamino-2-methyl-Oral LD₅₀ 74 mg/kg (rat) Inhalation 2- and 4-hour 1 -hour LC₅₀ ALC (rats) = (male rats) =(rat) 111 ppm 71 **ppm** 1 -hour LC₅₀ (female rats) = 104 ppm 1 -hour LC₅₀ (rats combined sexes) =107 ppm ALD = 30-100Dermal No Data (rabbit) mg/kg Slight to mild **Dermal** Slight Irritation **Eye Irritation** Death Severe **Dermal** No Data Not a sensitizer Sensitization

Table 5: Acute Mammalian Toxicity

A summary of the available data on repeated dose, developmental, and reproductive toxicity is shown in Table 6. Repeated administration of 2-amino-2-methylpropanenitrile to rats via inhalation for 2 weeks at vapor concentrations of 0, 1.4, 7.3, or 22 ppm produced neither deaths nor differences in body weights or clinical observations. In addition, no toxicologically significant changes in hematology, clinical chemistry, urine analysis, organ weight, gross observations, or microscopic observations were seen. The NOEL for the study was 22 ppm. 2-Amino-2,3-dimethylbutanenitrile was tested in a 28-day dermal study in rats at doses of 3, 10, and 30 mg/kg. Although increased thyroid weights were observed at all dose levels, no pathologic changes to account for this finding were observed. Based on skin irritation observed at ≥ 10 mg/kg, the NOEL was 3 mg/kg. However, the authors state that the intent of the repeated exposure dermal study was to assess systemic toxicity, and since no evidence of systemic toxicity was observed, the NOEL for systemic toxicity for the study was 30 mg/kg. No effects were observed in the reproductive organs (testes, epididymides, prostate, and seminal vesicle) of the male rats treated with 2-amino-2-methypropanenitrile for 2 weeks or in male and female rats (testes and uterus) treated with 2-amino-2,3-dimethylbutanenitrile for 28 days. Since no data are available regarding developmental or reproductive toxicity, a repeat

dose/developmental/reproductive toxicity screening test with 2-amino-2-methylpropanenitrile following OECD guideline 422 is recommended.

Table 6: Repeated Dose, Developmental, and Reproductive Toxicity

	Propanenitrile, 2- amino-2-methyl-	Butanenitrile, 2- amino-2-methyl-	
Repeated Dose Toxicity (NOAEL)	2-week inhalation (rats) NOEL =	N/A	「「「「「「「「」」」」 「「「」」 「「」」 「「」」 「「」」 「「」」
Developmental Toxicity	No Data	No Data	Argentalie.
Reproductive Toxicity	No Data	No Data	in the s

No information was found regarding genetic toxicity for 2-amino-2-methylpropanenitrile and 2-amino-2-methylbutanenitrile. 2-Amino-2,3-dimethylbutanenitrile was not mutagenic when tested in an Ames assay with *Salmonella typhimurium*, with and without exogenous metabolic activation. Since no data are available regarding the clastogenic effects of the category members or the analog chemical, a chromosome aberration study with 2-amino-2-methylpropanenitrile following OECD guideline 473 is recommended.

Table 7: Genetic Toxicity

	Propanenitrile, 2- amino-2-methyl-	Butanenitrile, 2-amino-2-methyl-	舞台ではないた。 Control toward Electronic Control
Mutagenic	No Data	No Data	
Clastogenic	No Data	No Data	

Human Exposure

2-Amino-2-methylpropanenitrile and 2-amino-2-methylbutanenitrile are DuPont-limited intermediates. These two aminoalkymitriles are manufactured at one DuPont plant and are shipped by DOT 412 tank truck to another DuPont facility for conversion into the corresponding (AIBN) and 2,2'azobis-(2-

methylbutyronitrile) (AMBN). The aminoalkymitriles are not sold to third parties and are not consigned to toll manufacturers for conversion to the final products.

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The aminoalkylnitriles are produced in a closed system using ammonia, the appropriate ketone (acetone or 2-butanone), and HCN. The aminoalkylnitriles are hard piped to dedicated storage tanks and stored under an ammonia blanket. Off-gases associated with the aminoalkylnitrile process are vented to a flare stack. Each batch is sampled during manufacture and each storage tank is sampled daily and before loading. All sampling is done using a closed system that utilizes a container with a septum seal on the top with a needle type injector to prevent human exposure to both liquid and vapors. Sample analysis is conducted in a ventilated laboratory hood. Each aminoalkylnitrile has a required percentage of excess ketone for product quality control. The excess ketone is used as a marker for potential exposure during personnel air monitoring at the manufacturing site, since it is more volatile than the corresponding aminoalkylnitrile.

During loading at the manufacturing site, the trailer and the storage tank are connected to form a closed system to prevent exposure. Flex hose is connected to the liquid valve on the trailer and the liquid is fed through an induction pipe to the bottom of the trailer. The aminoalkylmtrile in liquid form is pumped into the trailer and the vapor from the container is vented back, through a separate vent line, into the storage tank that is being emptied. Both lines are purged before disconnecting from the trailer. There is no operator exposure during the loading operation.

Safety equipment used depends on the task being performed. During routine monitoring of manufacturing operations, operators wear chemical goggles, a hard hat, and full-body Nomex® garments. In the course of laboratory work in a vented hood, safety glasses with sideshields and rubber gloves are worn. During loading operations at the manufacturing site, operators wear appropriate personal protective equipment to protect themselves from liquid and vapor contact while on the trailer. PPE consists of Nomex® clothing, hardhat, chemical splash goggles, HCN personal monitor, radio, and neoprene gloves. Safety showers, eyewash stations and self-contained breathing apparatus (SCBA) are available in close proximity to the operations area. All first breaks into equipment that cannot be confirmed as having been decontaminated require, at a minimum, the use of a full acid suit and self-contained breathing apparatus (SCBA), such as the Scott Air Pack or air-line respirators.

At the DuPont use site, aminoalkylnitrile tank trucks are close-dome unloaded under a nitrogen blanket, and may be vented to a flare as needed. The stainless steel storage tanks and associated piping are designed to code to contain the **aminoalkylnitrile**, and have redundant hi-hi level interlocks to prevent overfilling. The aminoalkylnitrile is pumped through an air stripper to remove excess ammonia. The air exiting the stripper is routed to a flare. The stripped liquid aminoalkylnitrile flows to a reactor below liquid level and is completely converted to the corresponding **Vazo**® product in the subsequent reaction.

During unloading of the aminoalkylnitrile tank trucks at the DuPont use site, operators wear personal protective equipment consisting of neoprene chemical gloves sealed to an acid suit, boots, acid hood, and air supplied positive pressure respirator. During sampling of aminoalkylnitriles, chemical gloves and chemical acid hood are required. Safety showers, eyewash stations and self-contained breathing apparatus (SCBA) are available in close proximity to the operations area.

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The DuPont Acceptable Exposure Limit (AEL) for acetone in 2-amino-2-methylpropanenitrile is 500 ppm as an 8- and 12-hour TWA (time-weighted average); the AEL for 2-butanone in 2-amino-2-methylbutanenitrile is 200 ppm as an 8- and 12-hour TWA. Air monitoring at the manufacturing site has shown that ketone concentrations are well below their respective AELs. At the use site, air monitoring is conducted for the aminoalkylnitriles per se. Levels of the aminoalkylnitriles measured in short term air monitoring during unloading operations have been consistently below 0.5 ppm, the limit of quantitation, and well below the DuPont AEL for 2-amino-2-methylpropanenitrile, which is 5 ppm (15-minute TWA). Results are shown in the table below:

Exposure Data:

Job Sampled	No. of Results	Average (ppm)	Minimum (ppm)	Maximum (ppm)
DuPont Manufacturing Site Operators (full shift) – acetone	6	<0.82	0.75	1.04
DuPont Manufacturing Site Operators (full shift) = 2-butanone	6	<0.81	0.67	1.56
DuPont End Use Site Operators during unloading of aminoalkylnitrile – as 2-amino-2-methylpropanenitrile or 2-amino-2-methylbutanenitrile	23	All < 0.5		

Conclusion

2-Amino-2-methylpropanenitrile and 2-amino-2-methylbutanenitrile may be considered as members of an HPV category based on the similarities in their molecular structures, reactivity, use, physical/chemical characteristics, and hazards. These two substances are nearest homologues and have the same functional groups. The use of supporting data **from** the next higher homologue, 2-amino-2,3-dimethylbutanenitrile, is consistent with the Agency's directive to HPV participants to maximize the use of scientifically appropriate data for related chemicals. Although some chemical and biological differences among these homologues may be expected, we believe these differences are minor. Generation of the additional data noted in the following test plan should be adequate to complete the HPV characterization of both members of the aminoalkylnitrile category.

Table 8: 2-Aminoalkylnitrile Category Proposed SIDS Test Plan

	Propanenitrile, 2-amino-2-methyl-	Butanenitrile, 2-amino-2-methyl-
Melting Point	Y	Y
Boiling Point	N ^a	N ^a
Vapor Pressure	Y ^b	Y ^b
Water Solubility	<u>c</u>	С
Stability in Water	Y	Y
Biodegradation	d	d
Repeated Dose/Reproductive/ Developmental Toxicity Screen	Y	N
Genetic Toxicity Chromosomal Aberrations	Y	N

^a The test samples will be observed for evidence of decomposition during melting point measurement.

Reference for Summary

Kirk-Othmer Encyclopedia of Chemical Technology (1978). 3rd edition, Wiley-Interscience.

^b The commitment to perform these studies is contingent upon the technical feasibility of running these studies.

^c Information to be determined in conjunction with testing of stability in water.

^d Need for performing these studies will be determined based on results of the stability in water testing.

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IUCLID

Data Set

Existing Chemical

CAS No.

: ID: 4475-95-0 : 4475-95-0

Substance name

: Butanenitrile, 2-amino-2-methyl-

EC No.

: 224-752-6

Molecular Formula

: C5H10N2

Producer related part

Company

: E. I. du Pont de Nemours and Company

Creation date

: 09.02.2006

Substance related part

Company

: E. I. du Pont de Nemours and Company

Creation date : 09.02.2006

Status Memo

:

11101110

: 02.06.2006

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Chapter (profile)
Reliability (profile)

: Chapter: 1, 2, 3, 4, 5, 6, 7, 8, 10 : Reliability: without reliability, 1, 2, 3, 4

Flags (profile)

: Flags: without flag, confidential, non confidential, WGK (DE), TA-Luft (DE), Material Safety Dataset, Risk Assessment, Directive 67/548/EEC, SIDS

1. General Information

ld 4475-95-0 **Date** 02.06.2006

1.0.1 APPLICANT AND COMPANY INFORMATION

1.0.2 LOGATION OF PRODUCTION SITE, IMPORTER OR FORMULATOR

1.0.3 IDENTITY OF RECIPIENTS

1.0.4 DETAILS ON CATEGORY/TEMPLATE

1.1.0 SUBSTANCE IDENTIFICATION

1.1.1 GENERAL SUBSTANCE INFORMATION

Purity type

Substance type

Physical status

Purity

Colour Odour liquid

: yellow

: ammonia-like odor

Attached document : 4475.bmp

$$H_3C$$
 H_2
 H_3C
 C
 C
 C
 C
 C
 C
 C
 C
 C

15.05.2006

1.1.2 SPECTRA

1.2 SYNONYMS AND TRADENAMES

1-Cyano-1-methylpropylamine

09.02.2006

2-Amino-2-methylbutanenitrile

09.02.2006

2-Amino-2-methylbutyronitrile

1. General Information

ld 4475-95-0 Date 02.06.2006

	Date 02.00.2000
09.02.2006	,
Isovalinonitrile	
09.02.2006	
Vazo 67 aminonitrile	
09.02.2006	
1.3 IMPURITIES	
1.4 ADDITIVES	
1.5 TOTAL QUANTITY	
1.6.1 LABELLING	
1.6.2 CLASSIFICATION	
1.6.3 PACKAGING	
1.7 USE PATTERN	
1.7.1 DETAILED USE PATTERN	
1.7.2 METHODS OF MANUFACTURE	
1.8 REGULATORY MEASURES	
1.8.1 OCCUPATIONAL EXPOSURE LI	MITVALUES
1.8.2 ACCEPTABLE RESIDUES LEVE	
1.83 WATER POLLUTION	
1.8.4 MAJOR ACCIDENT HAZARDS	

1. General Information **id** 4475-95-0 Date 02.06.2006 1.8.5 AIR POLLUTION 1.8.6 LISTINGS E.G. CHEMICAL INVENTORIES 1.9.1 DEGRADATION/TRANSFORMATION PRODUCTS 1.9.2 COMPONENTS 1.10 SOURGE OF EXPOSURE 1:11 ADDITIONAL REMARKS 1.12 LAST LITERATURE SEARCH 1.13 REVIEWS

ld 4475-95-0 Date 02.06.2006

2.1 MELTING POINT

Value

7.1 °C

Sublimation Method

Year **GLP**

Test substance

as prescribed by 1.1 - 1.4

Method

Modeled. MPBPWIN, v.1.41, module of EPIWIN 3.11 (Syracuse Research Corporation). MPBPWIN estimates melting point by 2 different methods. The first is an adaptation of the Joback group contribution method for melting point (Joback, 1982; Reid et al., 1987) and the second is a simple

Gold and Ogle method suggested by Lyman, 1985.

Remark

Reliability: Estimated value based on accepted model.

Result 09.02.2006 : Value at 760 mm Hg

(14) (16) (27)

Remark

: Additional Reference for Melting Point

17.02.2006

(9)

BOILING POINT

Decomposition

yes

Method

Year

no data

GLP Test substance

as prescribed by 1.1 - 1.4

Remark

15.05.2006

Reliability: Not assignable because limited study information was available.

2.3 DENSITY

Type Value relative density .886 at °C

Method

Year **GLP**

no data

Test substance

as prescribed by 1.1 - 1.4

Remark

Reliability: Not assignable because limited study information was available.

Result

Vapor density >1, where air = 1

15.05.2006

(9)

2.3.1 GRANULOMETRY

2.4 VAPOUR PRESSURE

Value

1.03 at 25 °C

Decomposition Method

other (calculated): MPBPWIN v1.40

5 / 25

ld 4475-95-0 Date 02.06.2006

Year **GLP**

no data

Test substance

as prescribed by 1.1 - 1.4

Method

: Estimated as the mean of Antoine & Grain methods

Syracuse Research Corporation (MPBPWIN v1.40 in EPIWIN v3.11 program) estimates the vapor pressure using the modified Grain method.

A description of the methodology is detailed in Lyman, 1985.

Remark

Reliability: Estimated value based on accepted model.

15.05.2006

Value

18.66513 hPa at 68 °C

Decomposition

Method Year

GLP

Test substance

as prescribed by 1.1 - 1.4

Remark

Estimated value based on accepted model.

Result

14 mmHg at 68 degree C (converted to 18.66513 hPa).

09.02.2006

(9)

(10)(16)

PARTITION COEFFICIENT

Partition coefficient

octanol-water

Log pow

-2.73 at 25 °C

pH value Method

other (calculated): KOWWIN, v1.67

Year

GLP

Test substance

as prescribed by 1.1 - 1.4

Method

Modeled. KOWWIN, v. 1.67, module of EPIWIN 3.11 (Syracuse Research Corporation). KOWWIN uses "fragment constant" methodologies to predict log P. In a "fragment constant" method, a structure is divided into

fragments (atom or larger functional groups) and coefficient values of each fragment or group are summed together to yield the log P estimate.

Remark

Reliability: Estimated value based on accepted model.

Test substance

(SMILES: C(#N)C(N(H)(H)(H)(CL))(C)CC) as ionized salt at environmental

pH and high dilution

09.02.2006

(10)(22)

Partition coefficient

octanol-water

Log pow pH value .45 at 25 °C

Method

other (calculated): KOWWIN, v1.67

Year

GLP

Test substance

as prescribed by 1.1 - 1.4

Method

Modeled. KOWWIN, v. 1.67, module of EPIWIN 3.11 (Syracuse Research Corporation). KOWWIN uses "fragment constant" methodologies to predict

log P. In a "fragment constant" method, a structure is divided into fragments (atom or larger functional groups) and coefficient values of each

fragment or group are summed together to yield the log P estimate.

Remark Test substance Reliability: Estimated value based on accepted model.

09.02.2006

(SMILES: C(#N)C(N)(C)CC)

(10)(22)

ld 4475-95-0 Date 02.06.2006

2.6.1 SOLUBILITY IN DIFFERENT MEDIA

Solubility in

Water

Value

27 g/l at °C

pH value

concentration

Temperature effects Examine different pol.

pKa

4.9 at 25 °C

at °C

Description Stable

Deg. product Method

other: calculated WSKOWIN v1.41

Year **GLP**

Test substance

as prescribed by 1.1 - 1.4

Method

Modeled.

Solubility - WSKOWWIN v.1.41, module of EPIWIN 3.11 (Syracuse Research Corporation). Water solubility is estimated from log Kow using

molecular weight and molecular fragment correction factors.

Modeled. pKa - SPARC On-line calculator, University of Georgia

Remark 15.05.2006 Reliability: Estimated value based on accepted model.

(24)(26)

(9)

Remark 15.05.2006 : Additional reference for water solubility

2.6.2 SURFACE TENSION

Value

1.7.°C

Type

Method

other: SFCC

Year **GLP**

no data

Test substance

as prescribed by 1.1 - 1.4

Remark

Reliability: Not assignable because limited study information was available.

Result

1.7°C (autodecomposition ~80°C)

17.02.2006

(9)

2.8 AUTO FLAMMABILITY

Result

flammable

Method Year

GLP

no data

ld 4475-95-0 **Date** 02.06.2006

Test substance	:	as prescri	bed by 1.1 -	- 1.4		÷			
Remark 17.02.2006	:	Reliability	: Not assign	able because	limited s	tudy inforn	nation wa	s availat	ole. (9)
2.10 EXPLOSIVE PROP	ERT	IES .							
2.11 OXIDIZING PROPE	ERTIE	S					建建场 的		
2.12 DISSOCIATION CO	DNST	ANT						(¥1
2.13 VISCOSITY									

2.14 ADDITIONAL REMARKS

3. Environmental Fate and Pathways

ld 4475-95-0 **Date** 02.06.2006

3.1.1 PHOTODEGRADATION

Deg. product

Method

other (calculated): AOPWIN v1.91

Year

GLP

: no

Test substance

: as prescribed by 1.1 - 1.4

Remark

Reliability: Estimated value based on accepted model.

Result

: Direct Photolysis: No Data

Indirect Photolysis: Estimated half-life = 10.8 days, due to OH radical oxidation in the atmosphere. With an estimated vapor pressure of 1.03 mm Hg (25 degree C) 2-amino-2-methylbutanenitrile will exist as a vapor in the

atmosphere.

Breakdown Products: No Data

15.05.2006

(10) (21)

3.1.2 STABILITY IN WATER

Deg. product

g. product

Method

other (calculated): HYDROWIN v1.67

Year

GLP

no

Test substance

: as prescribed by 1.1 - 1.4

Method

Modeled. HYDROWIN, v. 1.67 module of EPIWIN v3.11 (Syracuse Research Corporation). HYDROWIN cannot estimate a hydrolysis rate

constant for this type of chemical structure.

Remark

Reliability: Estimated value based on accepted model.

Result

% Hydrolyzed: No Data

Half-life: In the presence of water and the absence of excess ammonia, aminonitriles may disproportionate into their constituents: ketone, cyanide,

and ammonium (Kirk-Othmer, 1978)

15.05.2006

(15) (25)

3.1.3 STABILITY IN SOIL

3.2.1 MONITORING DATA

3.2.2. FIELD STUDIES

3.3.1 TRANSPORT BETWEEN ENVIRONMENTAL COMPARTMENTS

Type

fugacity model level III

Media Air : other: Air, Water, Soil, and Sediments

Water Soil Biota % (Fugacity Model Level I)% (Fugacity Model Level I)% (Fugacity Model Level I)

% (Fugacity Model Level II/III)

3. Environmental Fate and Pathways

ld 4475-95-0 Date 02.06.2006

Soil Method % (Fugacity Model Level II/III)

Year

Method

Modeled.

SMILES: C(#N)C(N)(C)CC Molecular Wt: 98.15

Vapor Pressure: 1.03 mm Hg (MPBPWIN program)

Log Kow: -0.25 (KOWWIN program)

other: Mackay, Level III Fugacity Model

Henry's Law Constant - HENRYWIN v. 3.10 module of EPIWIN v3.11 (Syracuse Research Corporation). Henry's Law Constant (HLC) is estimated by 2 separate methods that yield two separate estimates. The first method is the bond contribution method and the second is the group contribution method. The bond contribution method is able to estimate many more types of structures; however, the group method estimate is usually preferred (but not always) when all fragment values are available.

Koc - Calculated from log Kow by the Mackay Level III fugacity model incorporated into EPIWIN v3.11 (Syracuse Research Corporation).

Environmental Distribution - Mackay Level III fugacity model, in EPIWIN v3.11 (Syracuse Research Corporation). Emissions (1000 kg/hr) to air, water, and soil compartments.

Fugacity - The methodology and programming for the Level III fugacity model incorporated into EPIWIN v3.05 (Syracuse Research Corporation) were developed by Dr. Donald MacKay and coworkers and are detailed in

Mackay, 1991; Mackay et al. 1996a; Mackay et al. 1996b. Reliability: Estimated value based on accepted model.

Remark Result

Compartment % of total 1/2 life (hours)

distribution (advection + reaction) Air 696 0.1 Water 44.8 900

Soil 55 1800 Sediment 0.09 8100

Absorption Coefficient: Koc = 0.231 (calc by model)

Volatility: Henry's Law Constant = 7.35x10E-9 atm-m3/mole (HENRYWIN

program)

15.05.2006

(13) (17) (18) (19) (20)

3.3.2 DISTRIBUTION

MODE OF DEGRADATION IN ACTUAL USE

3.5 BIODEGRADATION

Deg. product

Method

other: Calculated BIOWIN v4.01

Year **GLP**

Test substance as prescribed by 1.1 - 1.4

10 / 25

3. Environmental Fate and Pathways

ld 4475-95-0 **Date** 02.06.2006

Method : Modeled. BIOWIN, v. 4.01 module of EPIWIN v3.11 (Syracuse Research

Corporation). BIOWIN estimates the probability for the rapid aerobic biodegradation of an organic chemical in the presence of mixed

populations of environmental microorganisms. Estimates are based upon fragment constants that were developed using multiple linear and non-

linear regression analyses.

Remark : Reliability: Estimated value based on accepted model.

Result : Linear Model Prediction: 0.9777 (Biodegrades Fast)

Non-Linear Model Prediction: 0.9965 (Biodegrades Fast)
Ultimate Biodegradation Timeframe: 2.7122 (weeks to months)
Primary Biodegradation Timeframe: 3.5308 (days to weeks)
MITI Linear Model Prediction: 0.06429 (readily degradable)
MITI Non-Linear Model Prediction: 0.6249 (readily degradable)

Breakdown Products: No Data

15.05.2006 (2) (10) (11) (12) (28)

3.6 BOD5, COD OR BOD5/COD RATIO

3.7 BIOACCUMULATION

Elimination

Method : other: calculated BCFWIN v2.15

Year

GLP : n

Test substance : as prescribed by 1.1 - 1.4

Method : Modeled. BCFWIN v. 2.15 module of EPIWIN v3.11 (Syracuse Research

Corporation). BCFWIN estimates the bioconcentration factor (BCF) of an organic compound using the compound's log octanol-water partition coefficient (Kow) with correction factors based on molecular fragments.

Remark : Reliability: Estimated value based on accepted model.

Result : log BCF = 0.5 (unionized or salt)

15.05.2006 (1)

3.8 ADDITIONAL REMARKS

ld 4475-95-0

Date 02.06.2006

4.1 ACUTE/PROLONGED TOXICITY TO FISH

Type

Species

Pimephales promelas (Fish, fresh water)

Exposure period Unit LC50

mg/l .71 other

Method Year GLP

1992 no

96 hour(s)

Test substance : as prescribed by 1.1 - 1.4

Method

No specific test guideline was reported; however, a scientifically defensible

approach was used to conduct the study.

A 96-hour unaerated, static, acute test using fathead minnows was performed at nominal concentrations of 0, 0.5, 1.0, 50, 500, and 5000 mg/L. Fish were <1 g at study start, and fish loading was <5 g per 4 L test solution. One test chamber per concentration with 5 animals per test chamber were used. The photoperiod was 16 hours light:8 hours dark. Dissolved oxygen and pH were measured in the 0, 0.5, 50, and 5000 mg/L nominal concentrations. No information regarding hardness, alkalinity, pH,

TOC, TSS, or salinity of the dilution water chemistry was reported.

Remark : Reliability: Medium because a suboptimal study design (nominal test

concentrations) was used.

Result: The LC50 was 0.71 mg/L (95% confidence limit, 0.5-1.0 mg/L).

Mortalities of 0, 0, 100, 100, 100, and 100% were observed at 0, 0.5, 1.0, 50, 500, and 5000 mg/L, respectively. Based on visual observations, the test substance was soluble in well water at all but the highest test concentration. At 5000 mg/L, a precipitate was seen after 24 hours. Temperature ranged from 20.7-22.2°C in the 0 mg/L group. The dissolved oxygen at 0 and 96 hours or at total mortality were 8.8, 8.9, 8.7, and 8.8 mg/L and 7.2, 5.8, 8.7, and 8.8, for the 0, 0.5, 1.0, 50, 500, and 5000 mg/L groups, respectively. The pH values at 0 and 96 hours or at total mortality were 6.8, 6.8, 8.6, and 9.5 and 6.9, 6.9, 8.6, and 9.5 for the 0, 0.5, 1.0, 50,

500, and 5000 mg/L groups, respectively. 2-Amino-2-methylbutanenitrile, purity 78%

Test substance

24.05.2006

(8)

Type

Species Exposure period

other: Fish 96 hour(s)

Unit LC50 mg/l 744.5

Method

other: ECOSAR v.0.993

Year GLP

: no

Test substance

as prescribed by 1.1 - 1.4

Remark Result Reliability: Estimated value based on accepted model.

744.5 mg/L; $\log \text{Kow} = -0.25$

15.05.2006

(23)

4.2 ACUTE TOXICITY TO AQUATIC INVERTEBRATES

Type

4. Ecotoxicity

ld 4475-95-0 Date 02.06.2006

Species

Exposure period

: Daphnia sp. (Crustacea) : 48 hour(s)

Unit EC50 : mg/l 41.1

Method

: other: ECOSAR v0.993

Year **GLP**

Test substance

: as prescribed by 1.1 - 1.4

Remark Result

: Reliability: Estimated value based on accepted model.

: 41.1 mg/L; $\log \text{Kow} = -0.25$

15.05.2006

(23)

4.3 TOXICITY TO AQUATIC PLANTS E.G. ALGAE

Species

other algae: green

Endpoint

Exposure period

Test substance

: 96 hour(s)

Unit

EC50 Method : = 35.7 calculated : other: Modeled

Year

GLP

: as prescribed by 1.1 - 1.4

Result

: 35.7 mg/L; $\log \text{Kow} = -0.25$

15.05.2006

(23)

4.4 TOXICITY TO MICROORGANISMS E.G. BACTERIA

4.5.1 CHRONIC TOXICITY TO FISH

4.5.2 CHRONIC TOXICITY TO AQUATIC INVERTEBRATES

4.6.1 TOXICITY TO SEDIMENT DWELLING ORGANISMS

4.6.2 TOXICITY TO TERRESTRIAL PLANTS

4.6.3 TOXICITY TO SOIL DWELLING ORGANISMS

4.6.4 TOX. TO OTHER NON MAMM. TERR. SPECIES

4.7 BIOLOGICAL EFFECTS MONITORING

4.8 BIOTRANSFORMATION AND KINETICS

4. Ecotoxicity

ld 4475-95-0 **Date** 02.06.2006

4.9 ADDITIONAL REMARKS

5.0 TOXICOKINETICS, METABOLISM AND DISTRIBUTION

5.1.1 ACUTE ORAL TOXICITY

Type : LD50

Value : 74 mg/kg bw

Species : rat

Strain : other: ChR-CD

Sex : male

Number of animals

Vehicle : other: corn oil

Doses : 70, 75, 80, and 90 mg/kg

Method : other: The method used is not specified

Year : 198 **GLP** : no

Test substance : as prescribed by 1.1 - 1.4

Method : No specific test guideline was reported; however, a scientifically defensible

approach was used to conduct the study.

The test substance, as a suspension in corn oil, was administered by gavage in single doses to 4 groups of 10 young adult rats. Dose levels were 70, 75, 80, and 90 mg/kg. The surviving rats were weighed and observed during a 14-day recovery period, and then sacrificed. The LD50 value was calculated from the mortality data using the method of D. J.

Finney.

Remark : Reliability: High because a scientifically defensible or guideline method was

used.

Result : Mortality was 1/10, 7/10, 9/10, and 10/10 at 70, 75, 80, and 90 mg/kg,

respectively. All deaths occurred within 1 day after dosing. At 70 mg/kg only slight initial weight loss was observed. At 75 mg/kg, lethargy, gasping, moribundity, and prostration were observed on the day of dosing. Slight weight loss was observed in 2 of the 3 survivors on the day after dosing. At 80 mg/kg, salivation, tremors, lethargy, moribundity, prostration, and weakness were observed on the day of dosing. Slight weight loss was observed in the 1 survivor on the day after dosing. At 90 mg/kg, tremors, convulsions, gasping, lethargy, and moribundity were observed on the day

of dosing.

LC50 = 74 mg/kg (95% confidence limites, 71-76 mg/kg): 2-Amino-2-methylbutanenitrile, purity 80%

Test substance

15.05.2006

(3)

5.1.2 ACUTE INHALATION TOXICITY

Type : LC50 **Value** : 107 ppm

Species : rat

Strain : other: Crl:CD®BR
Sex : male/female

Number of animals

Vehicle

Doses : 102, 106, 120, 126, and 225 ppm

Exposure time : 1 hour(s)

Method : other

Year : 1987

GLP : yes

5. Toxicity

ld 4475-95-0 Date 02.06.2006

Test substance

: as prescribed by 1.1 - 1.4

Method

No specific test guideline was reported; however, a scientifically defensible approach was used to conduct the study.

Groups of 5 male and 5 female rats were exposed for 1 hour to the test substance in air at mean vapor concentrations of 102, 106, 120, 126, and 225 ppm. Rats were exposed nose-only. Rats were weighed prior to exposure, and were observed for clinical signs of toxicity during exposure. Surviving rats were weighed and observed daily for 14 days. No pathological evaluations were performed.

Two analytical methods were used to measure the atmospheric concentration of the test substance. A gas chromatographic analysis was used to measure the atmospheric concentration of aminonitrile vapor (active ingredient). A colorimetric method was used to estimate the atmospheric concentration of ammonia. Chamber temperature, relative humidity, and chamber oxygen content were recorded.

Remark

Reliability: High because a scientifically defensible or guideline method was used.

Result

107 ppm (estimated for male and female rats combined) Chamber temperature ranged from 28-34°C, relative humidity ranged from 9-19%, and chamber oxygen content was 21%.

Mortality in male rats was 0/5, 3/5, 5/5, 3/5, and 5/5 at 102, 106, 120, 126, and 225 ppm, respectively. Mortality in female rats was 0/5, 5/5, 5/5, 5/5, and 5/5 at 102, 106, 120, 126, and 225 ppm, respectively. The majority of the deaths occurred during exposure, with remaining deaths occurring within 1 day of exposure. During exposure, rats in all groups had a red nasal discharge and a diminished response to sound. Rats that survived the exposure were lethargic or prostrate when released from the restrainers. Rats that survived the recovery period had no significant weight loss or adverse clinical signs.

The LC50 for male rats was 111 ppm. The LC50's for female rats and for both sexes combined could not be calculated due to the steep doseresponse observed. However, these LC50's were estimated to be 104 and 107 ppm, respectively. Estimated ammonia concentrations were well below those expected to cause death.

Test substance 15.05.2006

2-Amino-2-methylbutanenitrile, purity 74.9%

(7)

5.1.3 ACUTE DERMAL TOXICITY

5.1.4 ACUTE TOXICITY, OTHER ROUTES

5.2.1 SKIN IRRITATION

Species

rabbit

Concentration

.5 other: mL

Exposure Exposure time

24 hour(s)

Number of animals Vehicle

PDII Result Classification

5. Toxicity

ld 4475-95-0 Date 02.06.2006

Method Year

other 1980 no

GLP

Test substance

as prescribed by 1.1 - 1.4

Method

No specific test guideline was reported; however, a scientifically defensible approach was used to conduct the study.

Six rabbits were clipped free of hair on the trunk and lateral area, and placed in FDA-type stocks. Doses of 0.5 mL of the test substance were applied to intact skin under gauze squares. Rubber sheeting was then loosely wrapped around the trunk and secured with adhesive tape. After 24 hours, the rabbits were removed from the stocks, the patches taken off, and the reactions observed. Observations were also made at 48 hours.

Remark

Reliability: High because a scientifically defensible or guideline method was

used.

Result

At the 24-hour observation, slight erythema in 3/6 rabbits and no erythema in 3/6 rabbits was observed. At the 48-hour observation, slight erythema in 2/6 rabbits and no erythema in 4/6 rabbits was observed. No edema was

observed throughout the study.

Test substance 15.05.2006

2-Amino-2-methylbutanenitrile, purity 80%

(5)

Remark

Data from this additional source support the study results summarized above. This study was not chosen for detailed summarization because the data were not substantially additive to the database.

15.05.2006

5.2.2 EYE IRRITATION

Species

rabbit

Concentration

Dose

Exposure time

Comment

Number of animals

Vehicle

Result

Classification

Method Year

other 1980 no

GLP Test substance

as prescribed by 1.1 - 1.4

Method

One-tenth mL of undiluted test substance was placed into the right conjunctival sac of each of 2 male albino rabbits. After 20 seconds, 1 treated eye was washed with tap water for 1 minute. The treated eye of the other rabbit was not washed. Observations of the cornea, iris, and conjunctiva were made with a hand slit lamp at 1 and 4 hours, and at 1, 2, 3, 14, 21, 27, and 34 days. Fluor-i-strip® stain and a biomicroscope were

used at examinations after the day of treatment.

Remark

Reliability: High because a scientifically defensible or guideline method was

used.

Result

The test substance produced generalized moderate to severe corneal cloudiness with the development of pannus, moderate iritis, and severe conjunctivitis. Severe generalized cloudiness and moderate iritis persisted. The conjunctiva was normal at 27 days. An eye dosed with the test substance and promptly washed had a small area of transient slight corneal cloudiness and mild conjunctivitis with no iritic effects. The washed 5. Toxicity

ld 4475-95-0 Date 02.06.2006

eve was normal at 2 days.

Test substance

: 2-Amino-2-methylbutanenitrile, purity 80%

15.05.2006

(6)

5.3 SENSITIZATION

Type

Species

guinea pig

Number of animals

Vehicle

Result

Classification

Method

other: Modified Buehlor method

Year **GLP**

1980

Test substance

as prescribed by 1.1 - 1.4

Method

No specific test guideline was reported; however, the design is a

modification of the Buehler method.

The primary irritation test was conducted on 10 unexposed guinea pigs with 0.05 mL of a 100% (as received) solution and 10% solution of the test substance in dimethyl phthalate (DMP) on shaved intact shoulder skin. The induction phase for sensitization was a series of 4 sacral intradermal injections of 0.1 mL of a 1.0% solution in DMP, 1 each week beginning 2 days after the test for primary irritation. After a 13-day rest period, the test guinea pigs were challenged for sensitization with 0.05 mL of a 100% (as received) solution and a 10% solution of test substance in DMP on shaved, intact shoulder skin. At the same time 10 unexposed guinea pigs (controls)

of the same age received identical topical applications.

Remark

Reliability: High because a scientifically defensible or guideline method was

used.

Result

The test substance produced neither sensitization nor irritation in 10 male

guinea pigs.

Test substance

15.05.2006

2-Amino-2-methylbutanenitrile, purity 80%

(4)

5.4 REPEATED DOSE TOXICITY

5.5 GENETIC TOXICITY IN VITRO

5.6 GENETIC TOXICITY 'IN VIVO'

5.7 CARCINGENICITY

5.8.1 TOXICITY TO FERTILITY

5.8.2 DEVELOPMENTAL TOXICITY/TERATOGENICITY

5. Toxicity		4475-95-0 02.06.2006
5.8.3 TOXICITY TO REPRODUCTION, OTHER STUDIES	1 1/2004	
5.9 SPECIFIC INVESTIGATIONS		
5.10 EXPOSURE EXPERIENCE		(#1 <u>1</u>
5.11 ADDITIONAL REMARKS		The second secon

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6. Analyt. Meth. for Detection and Identification

ld 4475-95-0 **Date** 02.06.2006

6.1 ANALYTICAL METHODS

6.2 DETECTION AND IDENTIFICATION

7. Eff. Against Target Org. and Intended Uses **Id** 4475-95-0 Date 02.06.2006 7.1 FUNCTION 7.2 EFFECTS ON ORGANISMS TO BE CONTROLLED 7.3 ORGANISMS TO BE PROTECTED

	Date 02.06.2006
8.1 METHODS HANDLING AND STORING	
8.2 FIRE GUIDANCE	
8.3 EMERGENCY MEASURES	
8.4 POSSIB. OF RENDERING SUBST. HARMLESS	
8.5 WASTE MANAGEMENT	
8.6 SIDE-EFFECTS DETECTION	
8.7 SUBSTANCE REGISTERED AS DANGEROUS FOR GROUND WATER	
8.8 REACTIVITY TOWARDS CONTAINER MATERIAL	

ld 4475-95-0

8. Meas. Nec. to Prot. Man, Animals, Environment

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9. References Id 4475-95-0 Date 02.06.2006

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	10.	Summar	/ and	Evaluation	1
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ld 4475-95-0 Date 02.06.2006

- 10.1 END POINT SUMMARY
- 10.2 HAZARD SUMMARY
- 10.3 RISK ASSESSMENT

2006 JUH - 8 RH 9: 41

IUCLID

Data Set

Existing Chemical

CAS No.

: ID: 19355-69-2

Substance name

: 19355-69-2

: Propanenitrile, 2-amino-2-methyl-: 242-989-3

EC No. Molecular Formula

: C4H8N2

Producer related part

Company Creation date : E. I. du Pont de Nemours and Company

: 07.02.2006

Substance related part

Company Creation date : E. I. du Pont de Nemours and Company

: 07.02.2006

Status Memo

Printing date

: 02.06.2006

Revision date

Date of last update

: 24.05.2006

Number of pages

: 27

Chapter (profile) Reliability (profile) Flags (profile)

: Chapter: 1, 2, 3, 4, 5, 6, 7, 8, 10 : Reliability: without reliability, 1, 2, 3, 4

: Flags: without flag, confidential, non confidential, WGK (DE), TA-Luft (DE), Material Safety Dataset, Risk Assessment, Directive 67/548/EEC, SIDS

1. General Information

ld 19355-69-2 **Date** 02.06.2006

1.0.1 APPLICANT AND COMPANY INFORMATION

1.0.2 LOCATION OF PRODUCTION SITE, IMPORTER OR FORMULATOR

1.0.3 IDENTITY OF RECIPIENTS

1.0.4 DETAILS ON CATEGORY/TEMPLATE

1.1.0 SUBSTANCE IDENTIFICATION

1.1.1 GENERAL SUBSTANCE INFORMATION

Purity type

Substance type Physical status

Purity

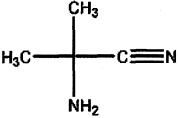
Colour Odour liquid

: brown

: ammonia-like

Attached document

: 19355-2.bmp



02.05.2006

1.1.2 SPECTRA

1.2 SYNONYMS AND TRADENAMES

1-Cyano-1-methylethylamine

07.02.2006

2-Amino-2-cyanopropane

07.02.2006

2-Amino-2-methylpropanenitrile

1. General Information

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	07.02.2006
	2-Amino-2-methylpropionitrile
	07.02.2006
	2-Aminoisobutyronitrile
	07.02.2006
	2-Aminopropane-2-carbonitrile
	15.05.2006
	2-Cyanoisopropylamine
	07.02.2006
	ABN
	07.02.2006
	ACAN
	07.02.2006
	alpha-Amino-alpha-methylpropionitrile
	07.02.2006
	alpha-Aminoisobutyronitrile
	07.02.2006
	Aminodimnethylacetonitrile
	07.02.2006
	Vazo 64AN
	07.02.2006
1	3 IMPURITIES - La
1	A ADDITIVES
1	5 TOTAL QUANTITY
1	.6.1 LABELLING
1	6.2 CLASSIFICATION

	Date 02.06.2006
 1.6.3 PACKAGING	
1.7 USE PATTERN	
1.7.1 DETAILED USE PATTERN	
1.7.2 METHODS OF MANUFACTUR	RE .
1.8 REGULATORY MEASURES	
15.05.2006	
1.8.1 OCCUPATIONAL EXPOSURE	LIMIT VALUES
Limit value : Short term exposure limit value Limit value : .0005	inute(s)
Result : 5 ppr 15.05.2006	n (15-minute TWA)
1.8.2 ACCEPTABLE RESIDUES LE	VELS TO THE REPORT OF THE PARTY
1.8.3 WATER POLLUTION	
1.8.4 MAJOR ACCIDENT HAZARD	
1.8.5 AIR POLLUTION	
1.8.6 LISTINGS E.G. CHEMICAL IN	VENTORIES
1.9.1 DEGRADATION/TRANSFORM	MATION PRODUCTS
1.9.2 COMPONENTS	

ld 19355-69-2

1. General Information

1. General Information

ld 19355-69-2 **Date** 02.06.2006

1.10 SOURCE OF EXPOSURE

1.11 ADDITIONAL REMARKS

Remark

: Existing published and unpublished data were collected and scientifically evaluated to determine the best possible study or studies to be summarized for each required endpoint. In the spirit of this voluntary program, other data of equal or lesser quality are not summarized, but are listed as additional references at the end of each appropriate section, with a statement to reflect the reason why these studies were not summarized.

07.02.2006

1.12 LAST LITERATURE SEARCH

1.13 REVIEWS

ld 19355-69-2 Date 02.06.2006

MELTING POINT 2.1

Value

-4.7 °C

Sublimation

Method

other: Modeled

Year

GLP

Test substance

as prescribed by 1.1 - 1.4

Method

Modeled. MPBPWIN, v.1.41, module of EPIWIN 3.11 (Syracuse Research Corporation). MPBPWIN estimates melting point by 2 different methods. The first is an adaptation of the Joback group contribution method for melting point (Joback, 1982; Reid et al., 1987) and the second is a simple Gold and Ogle method suggested by Lyman, 1985.

Remark

Reliability: Estimated value based on accepted model.

Result

07.02.2006

Value at 760 mmHg

(19) (21) (31)

Remark

: Additional Reference for Melting Point

07.02.2006

(12)

2.2 BOILING POINT

Decomposition

yes

Method

Year **GLP**

no data

Test substance

as prescribed by 1.1 - 1.4

Remark

08.02.2006

Reliability: Not assignable because limited study information was available.

Remark

Additional Reference for Boiling Point:

08.02.2006

(6)

2.3 DENSITY

Type Value relative density

Method

.9 at 25 °C

Year **GLP**

no data

Test substance

as prescribed by 1.1 - 1.4

Remark

08.02.2006

Reliability: Not assignable because limited study information was available.

(12)

Remark

08.02.2006

: Additional Reference for Density:

(6)

ld 19355-69-2 Date 02.06.2006

2.3.1 GRANULOMETRY

2.4 VAPOUR PRESSURE

Value

2.84 at 25 °C

Decomposition

other (calculated): MPBPWIN v1.40

Method Year

no

GLP Test substance

as prescribed by 1.1 - 1.4

Method

Estimated as the mean of Antoine & Grain methods

Syracuse Research Corporation (MPBPWIN) program in EPIWIN v3.11 estimates the vapor pressure using the modified Grain method. A

description of the methodology is detailed in Lyman, 1985.

Remark

15.05.2006

Reliability: Estimated value based on accepted model.

(14)(21)

Value

39.99 hPa at 66 °C

Decomposition

Method

Year **GLP**

no data

Test substance

: as prescribed by 1.1 - 1.4

Remark Result

Reliability: Not assignable because limited study information was available.

30 mmHg at 66 degree C (Converted to 39.99 hPa)

08.02.2006

(12)

Remark

: Additional Reference for Vapor Pressure:

08.02.2006

(6)

2.5 PARTITION COEFFICIENT

Partition coefficient

octanol-water

Log pow pH value

-3.23 at 25 °C

Method

other (calculated): KOWWIN, v1.67

Year

GLP

Test substance

as prescribed by 1.1 - 1.4

Method

Modeled. KOWWIN, v. 1.67, module of EPIWIN 3.11 (Syracuse Research Corporation). KOWWIN uses "fragment constant" methodologies to predict log P. In a "fragment constant" method, a structure is divided into

fragments (atom or larger functional groups) and coefficient values of each fragment or group are summed together to yield the log P estimate.

Remark

Reliability: Estimated value based on accepted model.

Test substance

(SMILES: C(#N)C(N(H)(H)(H)(CL))(C)C) as ionized salt at acidic

environmental pH and high dilution

08.02.2006

(14)(28)

Partition coefficient

Log pow

octanol-water -.04 at 25 °C

pH value

7 / 27

ld 19355-69-2 Date 02.06.2006

Method

other (calculated): KOWWIN, v1.67

Year

GLP

Test substance

as prescribed by 1.1 - 1.4

Method

Modeled. KOWWIN, v. 1.67, module of EPIWIN 3.11 (Syracuse Research Corporation). KOWWIN uses "fragment constant" methodologies to predict log P. In a "fragment constant" method, a structure is divided into

fragments (atom or larger functional groups) and coefficient values of each

fragment or group are summed together to yield the log P estimate.

Remark

Reliability: Estimated value based on accepted model.

Test substance

SMILES: C(#N)C(N)(C)C

08.02.2006

(14)(28)

(17)(25)

2.6.1 SOLUBILITY IN DIFFERENT MEDIA

Solubility in

Water

Value

> 100 g/l at °C

pH value

concentration

at °C

Temperature effects

Examine different pol.

pKa

4.9 at 25 °C

Description

Stable

Deg. product

Method

other: Modeled

Year

GLP

Test substance

as prescribed by 1.1 - 1.4

Method

pKa - SPARC On-line calculator, University of Georgia

Solubility - WSKOWWIN v.1.41, module of EPIWIN 3.11 (Syracuse Research Corporation). Water solubility is estimated from log Kow using

molecular weight and molecular fragment correction factors.

Remark

Reliability: Estimated value based on accepted model.

15.05.2006

Deg. product

Method Year **GLP**

Test substance

as prescribed by 1.1 - 1.4

Remark

Additional References for Water Solubility:

15.02.2006

(6)(12)

2.6.2 SURFACE TENSION

2.7 FLASH POINT

Value

1.7 °C

Type

Method

other: SFCC

Year **GLP**

no data

Test substance

as prescribed by 1.1 - 1.4

8 / 27

ld 19355-69-2 Date 02.06.2006

Remark 15.05.2006 : Reliability: Not assignable because limited study information was available.

AUTO FLAMMABILITY 2.8

2.9 FLAMMABILITY

Result

: flammable

Method

Year GLP

: no data

Test substance

: as prescribed by 1.1 - 1.4

Remark 15.05.2006 : Reliability: Not assignable because limited study information was available.

(12)

2.10 EXPLOSIVE PROPERTIES

2.11 OXIDIZING PROPERTIES

2.12 DISSOCIATION CONSTANT

2.13 VISCOSITY

2.14 ADDITIONAL REMARKS

ld 19355-69-2 Date 02.06.2006

3.1.1 PHOTODEGRADATION

Deg. product Method ...

other (calculated): AOPWIN v1.91

Year

GLP

Test substance

as prescribed by 1.1 - 1.4

Remark

Reliability: Estimate based on known qualitative structure-activity

relationships.

Result

: Direct Photolysis: No Data

Indirect Photolysis: Estimated half-life = 48 days, due to OH radical oxidation in the atmosphere. With an estimated vapor pressure of 2.84 mm Hg (25 degree C) 2-amino-2-methylpropanenitrile will exist as a vapor in

the atmosphere.

Breakdown Products: No Data

15.05.2006

(14)(27)

3.1.2 STABILITY IN WATER

Deg. product

Method

other (calculated): HYDROWIN v1.67

Year

GLP

Test substance

: as prescribed by 1.1 - 1.4

Method

Modeled. HYDROWIN, v. 1.67 module of EPIWIN v3.11 (Syracuse Research Corporation). HYDROWIN cannot estimate a hydrolysis rate

constant for this type of chemical structure.

Remark

Reliability: Estimated value based on accepted model.

Result

% Hydrolyzed: No Data

Half Life: In the presence of water and the absence of excess ammonia, aminonitriles may disproportionate into their constituents: ketone, cyanide,

and ammonium (Kirk-Othmer, 1978)

15.05.2006

(14) (20) (30)

3.1.3 STABILITY IN SOIL

3.2.1 MONITORING DATA

3.2.2 FIELD STUDIES

3.3.1 TRANSPORT BETWEEN ENVIRONMENTAL COMPARTMENTS

Type

fugacity model level III

Media

other: Air, Water, Soil, and Sediments

Air Water Soil

% (Fugacity Model Level I) % (Fugacity Model Level I)

% (Fugacity Model Level I)

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Biota Soil : % (Fugacity Model Level II/III) : % (Fugacity Model Level II/III)

Method

: other: Mackay, Level III Fugacity Model

Year

.

Method

Modeled.

SMILES: C(#N)C(N)(C)C Molecular Wt: 84.12

Vapor Pressure: 2.84 mm Hg (MPBPWIN program)

Log Kow: -0.04 (KOWWIN program)

Henry's Law Constant - HENRYWIN v. 3.10 module of EPIWIN v3.11 (Syracuse Research Corporation). Henry's Law Constant (HLC) is estimated by two separate methods that yield two separate estimates. The first method is the bond contribution method and the second is the group contribution method. The bond contribution method is able to estimate many more types of structures; however, the group method estimate is usually preferred (but not always) when all fragment values are available.

Koc - Calculated from Kow by the Mackay Level III fugacity model incorporated into EPIWIN v3.11 (Syracuse Research Corporation).

Environmental Distribution - Mackay Level III fugacity model, in EPIWIN v3.11 (Syracuse Research Corporation). Emissions (1000 kg/hr) to air, water, and soil compartments

Fugacity - The methodology and programming for the Level III fugacity model incorporated into EPIWIN v3.11 (Syracuse Research Corporation) were developed by Dr. Donald MacKay and coworkers and are detailed in

Mackay, 1991; Mackay et al. 1996a; Mackay et al. 1996b.

Remark Result Reliability: Estimated value based on accepted model.

Compartment % of total 1/2 life (hours)
distribution (advection + reaction)
Air 1 1150

Air 1 1150
Water 45.9 900
Soil 53.9 1800
Sediment 0.089 8100

Absorption Coefficient: Koc = 0.374 (calc by model)

Volatility: Henry's Law Constant = 5.54x10E-9 atm-m3/mole (HENRYWIN

program)

09.02.2006

(18) (22) (23) (24) (26)

3.3.2 DISTRIBUTION

3.4 MODE OF DEGRADATION IN ACTUAL USE

3.5 BIODEGRADATION

Deg. product

Method

other: Calculated BIOWIN v4.01

Year

GLP

no

Test substance

as prescribed by 1.1 - 1.4

ld 19355-69-2 **Date** 02.06.2006

Method

: Modeled. BIOWIN, v. 4.01 module of EPINWIN v3.11 (Syracuse Research

Corporation). BIOWIN estimates the probability for the rapid aerobic biodegradation of an organic chemical in the presence of mixed

populations of environmental microorganisms. Estimates are based upon fragment constants that were developed using multiple linear and non-

linear regression analyses.

Remark Result : Reliability: Estimated value based on accepted model.

Linear Model Prediction: 0.9844 (Biodegrades Fast)
Non-Linear Model Prediction: 0.9971 (Biodegrades Fast)

Ultimate Biodegradation Timeframe: 2.7432 (weeks to months)
Primary Biodegradation Timeframe: 3.5510 (days to weeks)
MITI Linear Model Prediction: 0.6353 (readily degradable)
MITI Non-Linear Model Prediction: 0.6188 (readily degradable)

Breakdown Products: No Data

15.05.2006

(2) (14) (15) (16) (32)

3.6 BOD5, COD OR BOD5/COD RATIO

3.7 BIOACCUMULATION

Elimination

Method

.

other: Modeled BCFWIN 2.15

Year GLP

:

Test substance

as prescribed by 1.1 - 1.4

Method

: BCFWIN v. 2.15 module of EPINWIN v3.11 (Syracuse Research

Corporation). BCFWIN estimates the bioconcentration factor (BCF) of an organic compound using the compound's log octanol-water partition coefficient (Kow) with correction factors based on molecular fragments

EPINWIN v3.11

Remark

Reliability: Estimated value based on accepted model.

Result

: log BCF = 0.5 (unionized or salt)

16.05.2006

(1)

3.8 ADDITIONAL REMARKS

4. Ecotoxicity

ld 19355-69-2

Date 02.06.2006

4.1 ACUTE/PROLONGED TOXICITY TO FISH

Type **Species**

Pimephales promelas (Fish, fresh water)

Exposure period

96 hour(s) mg/l

Unit LC50

.71 other

Method Year **GLP**

1992 : no

Test substance

as prescribed by 1.1 - 1.4

Method

No specific test guideline was reported; however, a scientifically defensible

approach was used to conduct the study.

The test substance was tested in an unaerated, static acute test. Nominal concentrations of the test substance were 0, 0.5, 1.0, 50, 500, and 5000 mg/L. Fish were <1 g at study start, and fish loading was <5 g per 4 L test solution. One test chamber per concentration with 5 animals per test chamber were used. The photoperiod was 16 hours light:8 hours dark. Dissolved oxygen and pH were measured in the 0, 0.5, 50, and 5000 mg/L nominal concentrations. No information regarding hardness, alkalinity,

TOC, or TSS of the dilution water chemistry was reported.

Remark

Reliability: Medium because a suboptimal study design (nominal test

concentrations) was used.

Result

The LC50 was 0.71 mg/L (95% confidence limit, 0.5-1.0 mg/L).

Mortality was 0, 0, 100, 100, 100, 100% at 0, 0.5, 1.0, 50, 500, and 5000

mg/L, respectively.

Based on visual observations, the test substance was soluble in well water at all but the 5000 mg/L test concentration, which had a slightly cloudy appearance. Temperature ranged from 20.7-21.2°C in the 0 mg/L group. Dissovled oxygen concentration at 0 hours was 8.6, 8.7, 8.6, and 9.4 mg/L at 0, 0.5, 50, and 5000 mg/L, respectively. Dissovled oxygen concentration at 96 hours or total mortality was 7.0, 7.7, 8.6, and 9.4 mg/L at 0, 0.5, 50, and 5000 mg/L, respectively. The pH values at 0 hours were 7.0, 7.1, 8.7, and 9.4 at 0, 0.5, 50, and 5000 mg/L, respectivley. The pH values at 96 hours or total mortality were 7.3, 7.5, 8.7, and 9.4 at 0, 0.5, 50, and 5000

mg/L, respectively.

Test substance

: 2-Amino-2-methylpropanenitrile, purity 74%

24.05.2006

(8)

Type

Species

other: Fish 96 hour(s)

Exposure period Unit LC50

mg/l 468.3

Method Year

other: ECOSAR

GLP

no

Test substance

as prescribed by 1.1 - 1.4

Method

: Modeled

Remark Result

Reliability: Estimated value based on accepted model.

468.3 mg/L; $\log \text{Kow} = -0.04$

15.02,2006

(29)

4. Ecotoxicity

ld 19355-69-2 **Date** 02.06.2006

(9)

4.2 ACUTE TOXICITY TO AQUATIC INVERTEBRATES

Туре

Species : Daphnia magna (Crustacea)

 Exposure period
 : 48 hour(s)

 Unit
 : mg/l

 EC50
 : 7.1

Method

Year : 1998 GLP : no

Test substance : as prescribed by 1.1 - 1.4

Method : No specific test guideline was reported; however, a scientifically defensible

approach was used to conduct the study.

Nominal concentrations used were 0, 0.5, 1.0, 50, 500, and 5000 mg/L. The test chamber was covered with clean glass plates. Two test chambers per concentration were used, with 5 daphnid neonates per test chamber. Dissolved oxygen and pH were recorded at 0 and 48 hours, or at total immobility. Water chemistry parameters at study start included alkalinity of 79 mg/L as CaCO3, hardness of 79 mg/L as CaCO3, and conductivity of 190 umhos/cm. Photoperiod was 16 hours light:8 hours dark. No information regarding TOC or TSS of the dilution water chemistry was

reported.

: Reliability: Medium because a suboptimal study design (nominal test

concentrations) was used.

Result : The LC50 was 7.1 mg/L (95% confidence limit, 1-50 mg/L).

Based on visual observations, the water control solution and the 0.5, 1.0, and 50 mg/L test solutions were clear with no color throughout the study. The 500 and 5000 mg/L test solutions were slightly cloudy at test start. Immobilities were 0, 0, 0, 100, 100, and 100% in the 0, 0.5, 1.0, 50, 500, and 5000 mg/L test concentrations, respectively. Total immobility at 500 and 5000 mg/L was observed within 2 minutes of test start. Water quality parameters were within acceptable limits, except the pH values in the 50, 500, and 5000 mg/L test solutions at the test start, which were slightly above the maximum limit of 9.0 (range 9.1-9.6), and the dissolved oxygen values in some of the test chambers exceeded the maximum limit of 105% saturation at test temperatures (i.e., approximately 9.7 mg/L at 20°C). pH at test start was 7.6, 7.8, 7.8, 9.1, 9.3, and 9.6 at 0, 0.5, 1, 50, 500, and 5000 mg/L, respectively. pH at 48 hours was 7.8-7.9, 8.0, and 8.0 at 0. 0.5, and 1 mg/L, respectively. Dissolved oxygen at test start was 9.5-9.6, 10.0-10.1, 10.1, 10.2-10.3, 9.7-10.4, and 9.4-9.5 at 0, 0.5, 1, 50, 500, and 5000 mg/L, respectively. Dissolved oxygen at 48 hours or total mortality was 8.9-9.0, 8.6-8.7, 8.5-8.8, and 9.9-10.0 at 0, 0.5, 1, and 50 mg/L, respectively. Temperature at study start was 20.2, 19.9-20.0, 19.9, 19.9, 19.9, and 19.8-19.9°C at 0, 0.5, 1, 50, 500, and 5000 mg/L, respectively. Temperature at 48 hours was 19.5, 19.4-19.5, and 19.4°C at 0, 0.5, and 1 mg/L, respectively.

Test substance 22.05.2006

Remark

2-Amino-2-methylpropanenitrile, purity 73-75%

Туре

Species : Daphnia sp. (Crustacea)

 Exposure period
 : 48 hour(s)

 Unit
 : mg/l

 EC50
 : 26.6

Method : other: Modeled by ECOSAR v.0.993

Year

GLP : no da

Test substance : as prescribed by 1.1 - 1.4

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4. Ecotoxicity

ld 19355-69-2

Date 02.06.2006

Remark

: Reliability: Estimated value based on accepted model.

Result

: 26.6 mg/L; $\log \text{Kow} = -0.04$

15.05.2006

(29)

4.3 TOXICITY TO AQUATIC PLANTS E.G. ALGAE

Species

other algae

Endpoint

Exposure period Unit

: 96 hour(s) : mg/l

EC50

Method

24.8

Year

: other: Modeled by ECOSAR v.0.993

GLP

Test substance

: as prescribed by 1.1 - 1.4

Remark

: Reliability: Estimated value based on accepted model.

Result : 24.8 mg/L; $\log \text{Kow} = -0.04$

17.02.2006

(29)

4.4 TOXICITY TO MICROORGANISMS E.G. BACTERIA

4.5.1 CHRONIC TOXICITY TO FISH

4.5.2 CHRONIC TOXICITY TO AQUATIC INVERTEBRATES

4.6.1 TOXICITY TO SEDIMENT DWELLING ORGANISMS

4.6.2 TOXICITY TO TERRESTRIAL PLANTS

4.6.3 TOXICITY TO SOIL DWELLING ORGANISMS

4.6.4 TOX. TO OTHER NON MAMM. TERR. SPECIES

BIOLOGICAL EFFECTS MONITORING

4.8 BIOTRANSFORMATION AND KINETICS

4.9 ADDITIONAL REMARKS

5.0 TOXICOKINETICS, METABOLISM AND DISTRIBUTION

5.1.1 ACUTE ORAL TOXICITY

Type

LD50

Value

10 - 30 mg/kg bw

Species

rat

Strain

other:Charles River Albino

Sex

6.....

Number of animals

female

Vehicle

Doses

3, 10, 30, 100, 300, 1000, 3000, and 10,000 mg/kg.

Method Year other 1974

GLP

no data

Test substance

: as prescribed by 1.1 - 1.4

Method

: No specific test guideline was reported.

An acute oral toxicity study was conducted on female rats (1/dose level), administered undiluted at 0.1 and 1.0% (w/v) aqueous solutions of 3, 10, 30, 100, 300, 1000, 3000, and 10,000 mg/kg. Body weights and clinical signs were periodically recorded. Surviving rats were sacrificed after 14

days. Gross necropsy was performed on all rats.

Remark

Reliability: High because a scientifically defensible or guideline method was

used.

Result

Rats dosed with 3 or 10 mg/kg survived. Mortality occurred in rats dosed with 30 mg/kg or higher. Death occurred in 1 minute to 1 ½ hours. The surviving animals at 3 and 10 mg/kg both gained weight over the 14-day observation period. Reactions exhibited by the rats (dose levels 30-10,000 mg/kg) within seconds after oral intubation included hypoactivity, labored breathing, salivation, straub tail, rhinitis, muscular weakness, tremors, fibrillary action, and convulsions. No reactions were noted in animals dosed at the 3 and 10 mg/kg dose levels. Necropsy examination of the animals that died revealed hemorrhages in the gastrointestinal tracts and dark red lungs. Gastroenteritis was noted at sacrifice in the rat dosed with 3 mg/kg. No other gross pathologic alterations were noted in rats

sacrificed at the end of the 14-day observation period.

Test substance 15.05.2006

2-Amino-2-methylpropanenitrile, purity not reported

(4)

5.1.2 ACUTE INHALATION TOXICITY

Type Value other: ALC 71 ppm

Species

: rat

Strain

other: Crl:CDR(SD)BR

Sex

male

Number of animals

Vehicle Doses

icle

22, 32, 46, 65, 71, 74

Exposure time Method Year 4 hour(s) other 1998

GLP Test substance

as prescribed by 1.1 - 1.4

16 / 27

ld 19355-69-2 Date 02.06.2006

Method

: No specific test guideline was reported.

Groups of 6 male rats each were exposed for 4 hours, whole body to vapor atmospheres of 22, 32, 46, 65, 71, or 74 ppm. Another group of 6 male rats was exposed whole body for a single 2-hour period to 74 ppm. Rats were approximately 6 -10 weeks old and weighed between 198-365 g at the time of exposure.

Rats were observed for mortality and response to alerting stimuli during the exposure and observed for mortality and clinical signs of toxicity after exposure. During a 14-day post exposure period, all surviving rats were observed each day for mortality, and were weighed and observed for clinical signs of toxicity at regular intervals. Surviving rats were sacrificed without pathological examination.

Chamber atmospheres were generated by flash evaporation of the test substance in air.

During exposure, rats were placed within wire-mesh cages and exposed whole-body inside the exposure chamber.

The atmospheric concentration of the test substance was determined by gas chromatography at approximately 30-minute intervals during each exposure. Chamber airflow was set at the beginning of each exposure to achieve at least 12 air changes per hour. Chamber temperature was targeted at 22±2°C. Chamber relative humidity was targeted at 50±10%. Airflow, temperature, and relative humidity were monitored continually.

Remark

Reliability: High because a scientifically defensible or guideline method was used.

Result

Mortality was 0/6, 0/6, 0/12, 0/6, 6/6, and 5/6 at 22, 32, 46, 65, 71, and 74 ppm, respectively. In general, rats died within 1 day of exposure. Clinical signs of toxicity observed included labored breathing, lethargy, gasping, nasal discharge, and stained fur. No clinical signs of toxicity were observed at < 65 ppm. In general, clinical signs of toxicity were observed for 1-2 days after exposure and had resolved by test day 3.

Weight losses of 4-8% were observed in some groups within 1 day of exposure. Rats generally resumed normal weight gains for the remainder of the recovery period.

Test substance 15.05.2006

2-Amino-2-methylpropanenitrile, purity 73-75%

(11)

Remark

Data from these additional sources support the study results summarized above. These studies were not chosen for detailed summarization because the data were not substantially additive to the database.

15.05.2006

(6)(7)

5.1.3 ACUTE DERMAL TOXICITY

Type Value other: ALD

Species

30 - 100 mg/kg bw

rabbit

Strain

New Zealand white

Sex

male

Number of animals

Vehicle Doses

30, 100, 300, 1000, 3000 mg/kg

Method

other: The method used is not specified

ld 19355-69-2 Date 02.06.2006

Year

1974

GLP

no data

Test substance

as prescribed by 1.1 - 1.4

Method

No specific test guideline was reported; however, a scientifically defensible

approach was used to conduct the study.

The test substance was administered undiluted on the abraded skin of rabbits (1/dose level) at doses of 30, 100, 300, 1000, and 3000 mg/kg. Body weights were recorded on days 0, 7, and 14. Gross pathology was

performed on all rabbits.

Remark

Reliability: High because a scientifically defensible or guideline method was

used.

Result

Mortality was 0, 100, 100, 100, and 100% at 30, 100, 300, 1000, and 3000 mg/kg, respectively. Death occurred in 10 to 35 minutes. The surviving rabbit given 30 mg/kg gained weight throughout the 14-day observation period. Clinical signs observed at 30 mg/kg or greater included excitation, hypoactivity, dyspnea, ataxia, muscular weakness, mydriasis, and miosis, which lasted from 6 to 22 hours. Clinical signs observed at ?100 mg/kg included excitation, rapid respiration, hypoactivity, mydriasis, dyspnea,

ataxia, muscular weakness, and tonic convulsions.

Test substance

15.05.2006

2-Amino-2-methylpropanenitrile, purity 76.3%

(3)

5.1.4 ACUTE TOXICITY, OTHER ROUTES

52.1 SKIN IRRITATION

Species

rabbit

Concentration

30 other: mg/kg

Exposure

Exposure time

Number of animals

Vehicle

PDII Result

Classification

Method

Year **GLP** Test substance other 1974 no data

as prescribed by 1.1 - 1.4

Method

No specific test guideline was reported.

The test substance was administered undiluted on the abraded skin of 1 male New Zealand White rabbit at a dose of 30 mg/kg (equivalent to 0.07 g

of test substance) for a dermal toxicity study.

Remark

Reliability: Low because an inappropriate method was used. The volume of test substance was insufficient to accurately assess dermal irritation.

Result

The test substance was slightly to mildly irritating to the skin. Skin changes at 24 hours were characterized by barely perceptible to pale red erythema. At 7 and 14 days, barely perceptible to pale red erythema and slight

desquamation were observed at the site of contact.

Test substance

15.05.2006

2-Amino-2-methylpropanenitrile, purity 76.3%

(3)

5.2.2 EYE IRRITATION

ld 19355-69-2 Date 02.06.2006

Species

rabbit .1 other: mL

Concentration

Dose

Exposure time

Comment Number of animals

Vehicle

Result Classification

Method

other:No specific test guideline was reported

Year **GLP**

1974 : no data

Test substance

as prescribed by 1.1 - 1.4

Method

No specific test guideline was reported; however, a scientifically defensible

approach was used to conduct the study.

The test substance (0.1 mL) was instilled undiluted into the eyes of 3 rabbits. In addition, 1 rabbit was exposed to 0.01 mL of undiluted test substance, and the eye remained unwashed.

Remark

Reliability: High because a scientifically defensible or guideline method was

used.

Result

Three rabbits dosed with 0.1 mL of the test substance exhibited immediate salivation and convulsions, and died within 5 minutes after instillation of the test substance. The rabbit dosed with 0.01 mL exhibited salivation, muscular weakness, hypoactivity, and diarrhea following instillation of the

test substance, and was found dead by 72 hours after instillation. Comeal, iritic, and conjunctival scores were 20, 5, and 12, respectively at 1 and 24

hours following instillation.

Test substance 15.05.2006

2-amino-2-methylpropanenitrile, purity not reported.

(5)

5.3 SENSITIZATION

5.4 REPEATED DOSE TOXICITY

Type

Species Sex

rat

Strain

: male

Route of admin.

: other: Crl:CDR(SD)BR inhalation

Exposure period

2 weeks

Frequency of treatm.

6 hours/day, 5 days/week

Post exposure period

0, 1.4, 7.3, 22 ppm

Doses

NOAEL

Control group

Method

= .0022 %

Year

other: The method used is not specified 1998

GLP

ves

Test substance

as prescribed by 1.1 - 1.4

Method

No specific test guideline was reported; however, a scientifically defensible approach was used to conduct the study.

Four groups of male rats, 10 per exposure level, were exposed wholebody to mean vapor concentrations of 0, 1.4, 7.3, or 22 ppm of the test

substance 6 hours per day for a total of 9 exposures.

ld 19355-69-2 **Date** 02.06.2006

The atmospheric concentration of the test substance was determined by gas chromatography at approximately 30-minute intervals during exposure. The exposure chambers were also sampled for ammonia and hydrogen cyanide. These compounds are breakdown products of the test substance. Airflow, temperature, and relative humidity were monitored continually.

During the exposure and a 14-day recovery phase rats were weighed and observed each day for clinical signs of toxicity.

After the last exposure, blood and urine samples were collected for clinical analyses, and 5 rats per group were sacrificed for pathologic examination. At the end of the 14 day recovery period, blood and urine samples were again collected, and all surviving rats were sacrificed for pathologic examination. Fifteen hematologic and 17 clinical chemistry parameters were measured or calculated, and 10 urine parameters were measured or examined.

Five rats per group were necropsied on test day 12. After a 14-day recovery period, the remaining 5 rats from each group were similarly necropsied. During the necropsy, liver, kidneys, lungs, testes, and brain were weighed. All rats were given a complete gross examination and representative samples of approximately 38 tissues were saved for possible histopathological evaluation. All tissues from test day 12 rats in the control and high (0 and 22 ppm, respectively) concentration groups were examined microscopically. Liver, kidneys, lungs, larynx/pharynx, nose, testes, and gross lesions from test day 12 low and intermediate (1.4 and 7.3 ppm, respectively) concentration groups, and 14-day recovery control and 22 ppm concentration groups were examined microscopically.

Descriptive statistics were used to summarize experimental data. Mean body weights and body weight gains were statistically analyzed with a oneway analysis of variance (ANOVA). Pairwise comparisons between test and control groups (sexes separate) were made with the Dunnett's test. For clinical laboratory data, ANOVA and Bartlett's test were calculated for each sampling time. Dunnett's test was used to compare means from the control groups and each of the groups exposed to the test substance. When the results of the Bartlett's test were significant, the Kruskal-Wallis test was employed and the Mann-Whitney U test was used to compare means from the control groups and each of the groups exposed to the test substance. Mean final body weights and mean absolute and relative (to body and brain) organ weights were analyzed by ANOVA. When the value of the F-statistic for differences among groups was significant, pairwise comparisons between treated and control groups were made with Dunnett's test. Bartlett's test was used to test for homogeneity of variances.

Remark

Result

- Reliability: Medium because there was no demonstrated effect at the highest concentration tested.
- : Analytically determined mean vapor concentrations of the test substance for the 3 test chambers were 1.4, 7.3, and 22 ppm. No hydrogen cyanide was found in the 1.4 ppm or 7.3 ppm chambers, but approximately 0.9 ppm was found in the 22 ppm chamber. Low concentrations of ammonia were present in all test chambers (1.5-6 ppm). Concentrations of breakdown products found in the chambers were not considered toxicologically significant.

The mean relative humidity for the chambers was between 37 and 41%, the mean chamber temperatures were 25 or 26°C, and the oxygen concentrations were 21%.

No deaths were observed during the study. No differences in body weight or clinical observations were observed during the study. No toxicologically

ld 19355-69-2 **Date** 02.06.2006

important changes occurred in hematology, clinical chemistry, or urine analytical parameters. No test substance-related changes in organ weights, gross observation, or microscopic observations were observed at any exposure level tested.

Under the conditions of this study, the no-observed-effect level (NOEL) was 22 ppm.

Test substance 15.05.2006

: 2-Amino-2-methylpropanenitrile, purity 73-75%

(10)

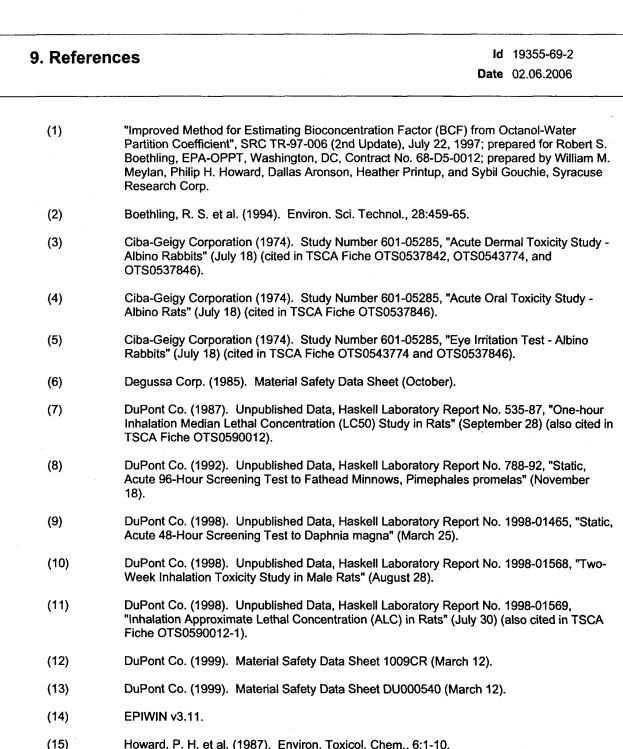
5.5 GENETIC TOXIC	ITY IN VITRO			
5.6 GENETIC TOXIC	TY IN VIVO		18	
5.7 CARCINOGENIC	lleY			
5.8.1 TOXICITY TO FE	RILLIY			
5.8.2 DEVELOPMENT	AL TOXICITY/TERA	TOGENICITY		
5.8.3 TOXICITY TO RE	PRODUCTION, OTH	IER STUDIES		
5.9 SPECIFIC INVES	TIGATIONS			
5.10 EXPOSURE EXP	PERIENCE			
		•		

5.11 ADDITIONAL REMARKS

6. Aı	nalyt. Meth. for Detection and Ide	entification	ld 19355-69-2 Date 02.06.2006
6.1	ANALYTICAL METHODS		
6.2	DETECTION AND IDENTIFICATION		
			,

7. Eff. Against Target Org. and Intended Uses		19355-69-2 02.06.2006
7.1 FUNCTION	11 - 5 - 11 - 11 - 11 - 11 - 11 - 11 -	
7.2 EFFECTS ON ORGANISMS TO BE CONTROLLED		
7.3 ORGANISMS TO BE PROTECTED		
7.4 USER		
7.5. RESISTANCE		
	,	

23 / 27



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- http://ibmlc2.chem.uga.edu/sparc/index.cfm (17)
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Tunkel, J. et al. (2000). Predicting Ready Biodegradability in the MITI Test. Environ.

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(32)

10. Summary and Evaluation	ld 19355-69-2 Date 02.06.2006
10.1 END POINT SUMMARY	
10.2 HAZARD SUMMARY	
10.3 RISK ASSESSMENT	

2006 JUN -8 AN 9:41

IUCLID

Data Set

Existing Chemical

: ID: 13893-53-3

CAS No.

: 13893-53-3

Substance name

: 2-Amino-2,3-dimethylbutanenitrile

Producer related part

Company

: E. I. du Pont de Nemours and Company

Creation date : 09.02.2006

Substance related part

Company

: E. I. du Pont de Nemours and Company

Creation date : 09.02.2006

Status Memo

Printing date

: 02.06.2006

Revision date

Date of last update

: 31.05.2006

Number of pages

: 28

Chapter (profile) Reliability (profile) : Chapter: 1, 2, 3, 4, 5, 6, 7, 8, 10

: Reliability: without reliability, 1, 2, 3, 4

Flags (profile) : Flags: without flag, confidential, non confidential, WGK (DE), TA-Luft (DE), Material Safety Dataset, Risk Assessment, Directive 67/548/EEC, SIDS

1. General Information

ld 13893-53-3 **Date** 02.06.2006

1.0.1 APPLICANT AND COMPANY INFORMATION

1.0.2 LOCATION OF PRODUCTION SITE, IMPORTER OR FORMULATOR

1.0.3 IDENTITY OF RECIPIENTS

1.0.4 DETAILS ON CATEGORY/TEMPLATE

1.1.0 SUBSTANCE IDENTIFICATION

1.1.1 GENERAL SUBSTANCE INFORMATION

Purity type
Substance type
Physical status
Purity

liquid

Purity Colour Odour

Attached document : untitled.bmp

11.05.2006

1.1.2 SPECTRA

1.2 SYNONYMS AND TRADENAMES

Aminonitrile

17.02.2006

1.3 IMPURITIES

1. General Inform	nation		13893-53-3 02.06.2006
1.4 ADDITIVES		《基础基本》。 《经理教》,1987年,198	
1.5 TOTAL QUANTI			1 - 1 - 1 - 1 - 1 - 1 - 1 - 1 - 1 - 1 -
1.6.1 LABELLING			
1.6.2 CLASSIFICATIO			
1.6.3 PACKAGING			
1.7 USE PATTERN			
1.7.1 DETAILED USE	PATTERN		The second secon
1.7.2 METHODS OF N	MANUFACTURE		
1.8 REGULATORY	MEASURES		
1.8.1 OCCUPATIONA	L EXPOSURE LIMIT VALUES		
Type of limit Limit value	: other: OSHA PEL (TWA) : 4.7 other: ppm		
Remark 11.05.2006	: 4.7 ppm (5 mg/m3): PEL/TLV		
1.8.2 ACCEPTABLE	RESIDUES LEVELS		
1.8.3 WATER POLLU	TION	2.02.35 Walliam 1997	
1.8.4 MAJOR ACCIDE	INT HAZAROS		
1.8.5 AIR POLLUTION			
1.8.6 LISTINGS E.G. (CHEMICAL INVENTORIES		

1. General Information

ld 13893-53-3 **Date** 02.06.2006

1.9.1 DEGRADATION/TRANSFORMATION PRODUCTS

1.9.2 COMPONENTS

1.10 SOURCE OF EXPOSURE

1.11 ADDITIONAL REMARKS

Remark

: Existing published and unpublished data were collected and scientifically evaluated to determine the best possible study or studies to be summarized for each required endpoint. In the spirit of this voluntary program, other data of equal or lesser quality are not summarized, but are listed as additional references at the end of each appropriate section, with a statement to reflect the reason why these studies were not summarized.

16.05.2006

1.12 LAST LITERATURE SEARCH

1.13 REVIEWS

ld 13893-53-3 **Date** 02.06.2006

2.1 MELTING POINT

Value

7.7 °C

Sublimation

. ,.,

Method

: other: MPBPWIN v1.41

Year

:

GLP Test substance

: as prescribed by 1.1 - 1.4

Method

: Modeled. MPBPWIN, v.1.41, module of EPIWIN 3.11 (Syracuse Research Corporation). MPBPWIN estimates melting point by 2 different methods. The first is an adaptation of the Joback group contribution method for melting point (Joback, 1982; Reid et al., 1987) and the second is a simple

Gold and Ogle method suggested by Lyman, 1985.

Remark

Reliability: Estimated value based on accepted model.

Result 17.02.2006

: Value at 760 mm Hg

(19) (20) (28)

Remark

: Additional Reference for Melting Point

17.02.2006

(14)

2.2 BOILING POINT

Value

186.9 °C at

Decomposition

:

Method

other:MPBPWIN Program v. 1.40

Year

GLP

no

Test substance

: as prescribed by 1.1 - 1.4

Method

Estimated by the MPBPWIN Program (v. 1.40), using the adapted Stein

and Brown Method.

Reliability

(2) valid with restrictions

Klimisch code: 2f

16.05.2006

(30)

2.3 DENSITY

2.3.1 GRANULOMETRY

2.4 VAPOUR PRESSURE

Value

.7999342 hPa at 25 °C

Decomposition

Test substance

.1999342 IIFa at 25

Method

other (calculated):Estimated by the MPBPWIN Program (v. 1.40),

Year

· no

GLP

: as prescribed by 1.1 - 1.4

Method

Estimated by the MPBPWIN Program (v. 1.40), using mean of Antoine and

Grain methods.

Result

0.6 mmHg at 25 degree C (converted into 0.7999342 hPa at 25 degree C

5/28

ld 13893-53-3 Date 02.06.2006

Reliability

(2) valid with restrictions

Klimisch code: 2f

17.02.2006

(30)

Value

31,2241 hPa at 25 °C

Decomposition Method

other (calculated): Modeled

Year

GLP Test substance

as prescribed by 1.1 - 1.4

Method

The vapor pressure was measured using a static method. The sample was placed in a glass cell and degassed using five freeze-pump-thaw cycles. The sample temperature was measured to ±0.01°C with a Hewlett-Packard Quartz Thermometer and controlled to ±1°C with a Blue-M forced air oven. The pressure was measured with a MKS Baratron capacitance transducer. The sample was stable during the experiment with no discoloration and it gave stable pressure reading once thermal equilibration was achieved.

Result

23.42 mmHg at 25 degree C (converted to 31.2241 hPa at 25 degree C

Reliability

(2) valid with restrictions

Klimisch code: 2e. This study was not conducted under GLP or OECD guidelines, but generally meets scientific standards, is well documented,

and is accepted for assessment.

17.02.2006

(11)

2.5 PARTITION COEFFICIENT

Partition coefficient

octanol-water -2.32 at °C

Log pow pH value

Method

other (calculated): KOWWIN, v1.67

Year **GLP**

Test substance

as prescribed by 1.1 - 1.4

Method

Modeled. KOWWIN, v.1.67, module of EPIWIN 3.11 (Syracuse Research Corporation). KOWWIN uses "fragment constant" methodologies to predict log P. In a "fragment constant" method, a structure is divided into fragments (atom or larger functional groups) and coefficient values of each fragment or group are summed together to yield the log P estimate.

Remark

Reliability: Estimated value based on accepted model.

Test substance

(SMILES: C(#N)C(N(H)(H)(H)(CL))(C(C)C)C) as ionized salt at acidic

environmental pH and high dilution

17.02.2006

(25)

Partition coefficient

octanol-water .87 at °C

Log pow pH value

other (calculated): KOWWIN, v1.67

Method Year **GLP**

Test substance

as prescribed by 1.1 - 1.4

Method

Remark

Modeled. KOWWIN, v.1.67, module of EPIWIN 3.11 (Syracuse Research Corporation). KOWWIN uses "fragment constant" methodologies to predict

log P. In a "fragment constant" method, a structure is divided into

fragments (atom or larger functional groups) and coefficient values of each

fragment or group are summed together to yield the log P estimate.

Reliability: Estimated value based on accepted model.

Test substance

SMILES: C(#N)C(N)(C(C)C)C)

ld 13893-53-3 Date 02.06.2006

(25)17.02.2006

2.6.1 SOLUBILITY IN DIFFERENT MEDIA

Solubility in

: Water

Value

107000 mg/l at 25 °C

pH value

concentration

: at °C

Temperature effects

Examine different pol. :

pKa

: 4.9 at 25 °C

Description

Stable

Year

Deg. product

Method

GLP

: as prescribed by 1.1 - 1.4

: other: WSKOW v1.40

Test substance

: Estimated from Kow with WSKOW (v1.40): KowWin Estimate

Method Reliability

: (2) valid with restrictions Klimisch code: 2f

11.05.2006

(27)(30)

2.6.2 SURFACE TENSION

2.7 FLASH POINT

2.8 AUTO FLAMMABILITY

2.9 FLAMMABILITY

2.10 EXPLOSIVE PROPERTIES

2.11 OXIDIZING PROPERTIES

2.12 DISSOCIATION CONSTANT

2.13 VISCOSITY

2.14 ADDITIONAL REMARKS

ld 13893-53-3 Date 02.06.2006

3.1.1 PHOTODEGRADATION

INDIRECT PHOTOLYSIS

Sensitizer

Conc. of sensitizer

Rate constant Degradation

Deg. product

Method

Year

GLP

Test substance

Method

Result

Reliability

as prescribed by 1.1 - 1.4

% after

Estimated by the AOP program (v1.90), which estimates rate constants

and half-lives of atmospheric reactions of organic compounds with hydroxyl

radicals and ozone in the atmosphere.

.000000000002888 cm3/(molecule*sec)

other (calculated): AOPWIN v1.90

Indirect Photolysis: For reaction with hydroxyl radicals, the predicted half-

life of the chemical is relatively rapid.

Rate constant: 2.888x10-12 cm3/molecule-sec

Half-life: 44,443 hours (2) valid with restrictions

Klimisch code: 2f

16.05.2006 (30)

3.1.2 STABILITY IN WATER

Deg. product

Method

other (calculated): HYDROWIN v1.67

Year

GLP

Test substance

as prescribed by 1.1 - 1.4

Remark

Haftlife: No Estimate available

Reliability: Estimated value based on accepted model.

Result % Hydrolyzed: The program was not able to estimate a hydrolysis rate

constant for this type of chemical structure. However, as manufactured, 2amino-2,3-dimethylbutanenitrile is prepared as an 80% solution in toluene, and this solution will partially hydrolyze in water by producing CN-, which will be detectable immediately. A small fraction of the 2-amino-2,3dimethylbutanenitrile dissociates under ambient conditions, whether as neat (100%) liquid or in solution with non-reactive organic solvents such as

toluene. CN- is a product of the dissociation of 2-amino-2,3dimethylbutanenitrile and will be present in a low concentration in equilibrium with 2-amino-2,3-dimethylbutanenitrile under all expected

conditions.

Aqueous wastes containing 2-amino-2,3-dimethylbutanenitrile, when commingled with a waste stream that is maintained at a pH of at least 10 by the addition of caustic, chemically decomposes the 2-amino-2,3dimethylbutanenitrile to CN-, ammonia, and methyl isopropyl ketone. Thus

indicating that with pH increase the material decomposes.

11.05.2006

(30)

3.1.3 STABILITY IN SOIL

ld 13893-53-3

Date 02.06.2006

3.2.1 MONITORING DATA

3.2.2 FIELD STUDIES

3.3.1 TRANSPORT BETWEEN ENVIRONMENTAL COMPARTMENTS

Modeled.

Type

fugacity model level III

Media

other: Air, Water, Soil, and Sediments

Air Water Soil

% (Fugacity Model Level I) % (Fugacity Model Level I)

% (Fugacity Model Level I) % (Fugacity Model Level II/III)

Biota Soil

% (Fugacity Model Level II/III)

other: (calculation) Level III Fugacity Model

Method Year

Method

: Fugacity - The methodology and programming for the Level III fugacity model incorporated into EPIWIN v3.05 (Syracuse Research Corporation) were developed by Dr. Donald MacKay and coworkers and are detailed in: **HENRYWIN -**

Henry's Law Constant - HENRYWIN v. 3.10 module of EPIWIN v3.11 (Syracuse Research Corporation). Henry's Law Constant (HLC) is estimated by two separate methods that yield two separate estimates. The first method is the bond contribution method and the second is the group contribution method. The bond contribution method is able to estimate many more types of structures; however, the group method estimate is usually preferred (but not always) when all fragment values are available.

Koc - Calculated from Kow by the Mackay Level III fugacity model incorporated into EPIWIN v3.11 (Syracuse Research Corporation).

Environmental Distribution - Mackay Level III fugacity model, in EPIWIN v3.11 (Syracuse Research Corporation). Emissions (1000 kg/hr) to air. water, and soil compartments.

Remark Result

Reliability: Estimated value based on accepted model.

Compartment % of total

1/2 life (hours)

distribution Air 0.127

(advection + reaction) 133

Water 42.2 Soil 57.6 Sediment 0.087 900 1800 8100

Absorption Coefficient: Koc = 3.04 (calc by model)

Volatility: Henry's Law Constant = 9.76x10E-9 atm-m3/mole (HENRYWIN program)

11.05.2006

(18) (21) (22) (23) (24)

Remark

Data from this additional source supports the study results summarized above. This study was not chosen for detailed summarization because the data were not substantially additive to the database.

16.05.2006

(15)

3. Environmental Fate and Pathways

ld 13893-53-3 **Date** 02.06.2006

3.3.2 DISTRIBUTION

3.4 MODE OF DEGRADATION IN ACTUAL USE

3.5 BIODEGRADATION

Deg. product

L.

Method

other: Calculated BIOWIN v4.01

Year GLP

. .

Test substance

: as prescribed by 1.1 - 1.4

Method

: Modeled. BIOWIN, v. 4.01 module of EPIWIN v3.11 (Syracuse Research Corporation). BIOWIN estimates the probability for the rapid aerobic biodegradation of an organic chemical in the presence of mixed

populations of environmental microorganisms. Estimates are based upon fragment constants that were developed using multiple linear and non-

linear regression analyses.

Remark Result Reliability: Estimated value based on accepted model.
 Linear Model Prediction: 0.9710 (Biodegrades Fast)
 Non-Linear Model Prediction: 0.9957 (Biodegrades Fast)

Ultimate Biodegradation Timeframe: 2.6812 (weeks-months)
Primary Biodegradation Timeframe: 3.5105 (days-weeks)
MITI Linear Model Prediction: 0.5015 (readily degradable)

MITI Non-Linear Model Prediction: 0.3999 (not readily degradable)

11.05.2006

(13) (16) (17) (31)

Remark

Data from this additional source supports the study results summarized above. This study was not chosen for detailed summarization because the

data were not substantially additive to the database.

16.05.2006

(29)

3.6 BOD5, COD OR BOD5/COD RATIO

3.7 BIOACCUMULATION

Elimination

Method

other: Modeled BCFWIN 2.15

Year GLP

: no

Test substance

: as prescribed by 1.1 - 1.4

Method

Modeled. BCFWIN v. 2.15 module of EPIWIN v3.11 (Syracuse Research Corporation). BCFWIN estimates the bioconcentration factor (BCF) of an organic compound using the compound's log octanol-water partition coefficient (Kow) with correction factors based on molecular fragments.

Remark

: Reliability: Estimated value based on accepted model.

Result

log BCF = 0.5 (unionized or salt)

16.05.2006

(1)

3.8 ADDITIONAL REMARKS

ld 13893-53-3

Date 02.06.2006

4.1 ACUTE/PROLONGED TOXICITY TO FISH

Type

static

Species

Lepomis macrochirus (Fish, fresh water)

Exposure period

96 hour(s)

Unit LC50 mg/l .75

Method

other: EPA 660-3-75-009

Year GLP 1984

Test substance

as prescribed by 1.1 - 1.4

Method

Patterned after EPA-660-3-75-009. ABC Laboratories Protocol 7601 (American Cyanamid Protocol 981-83-140).

The static fish bioassay was conducted in 5-gallon glass vessels containing 15 liters of soft reconstituted water. Ten fish with a mean weight of 0.34 g and a mean length of 25 mm were used for each test concentration. The test vessels were kept in a water bath at 22±1°C. A 48-hour range-finding test was conducted to determine the concentration range for the definitive study. The preliminary test concentrations were set at 0.1, 1.0, and 10 mg/L. Based on the results of the preliminary testing, 5 test concentrations were selected, 0.10, 0.18, 0.32, 0.56, and 1.0 mg/L. Exposures were based on nominal concentrations. Test concentrations were prepared by preparing a stock solution in deionized water and serially diluting to obtain

desired concentrations. All results were based on the nominal

concentrations.

The bluegill sunfish were challenged with a reference compound, Antimycin A, to verify that the fish were in good condition. The 96-hour LC50 for bluegill sunfish exposed to the control substance was 1.2x10E-4 mg/L, which indicates that the fish were in good condition.

The fish were observed once every 24 hours for mortality and abnormal effects. Water quality parameters of temperature, dissolved oxygen, and pH were measured throughout the test and were within acceptable limits.

Statistical analysis of the concentration versus effect data was obtained by employing a computerized LC50 program developed by Stephan. This program calculated the LC50 statistic and its 95% confidence limits using the binomial and the moving average tests, respectively. The method of calculation selected for use was that which gave the narrowest confidence

limits for the LC50.

Result

The no-effect concentration for the test material, based on the lack of mortality and abnormal effects was estimated to be 0.5 mg/L after 96 hours. All the fish in the 1.0 mg/L test concentration died on or before the 24-hour observation period. Water quality parameters of temperature, dissolved oxygen, and pH were measured throughout the test and were within acceptable limits.

Test substance Reliability 2-Amino-2,3-dimethylbutanenitrile, purity not reported

(2) valid with restrictions

Klimisch code: 2c

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(5)

Type

Species
Exposure period

other: Fish 96 hour(s)

Unit LC50 : mg/l : 163.3

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(26)

Method

other: ECOSAR v.0.993

Year

GLP

Test substance

as prescribed by 1.1 - 1.4

Remark

Reliability: Estimated value based on accepted model.

Result

163.3 mg/L; $\log \text{Kow} = 0.87$

11.05.2006

4.2 ACUTE TOXICITY TO AQUATIC INVERTEBRATES

Type

Species

Daphnia magna (Crustacea)

Exposure period

48 hour(s) mg/l

Unit

6.9

EC50 Method

other: US EPA 600/3-75009

Year

1984

GLP

Test substance

as prescribed by 1.1 - 1.4

Method

Based on methods outlined in the Committee on Methods for Toxicity Test with Aquatic Organisms, USEPA 600/3-75009. ABC Laboratories Protocol 7806 (American Cyanamid Protocol 981-83-137).

The static Daphnia magna bioassay was conducted in 250 mL glass beakers, 10 daphnids/beaker, containing 200 mL of ABC well water. These vessels were kept at 20±2°C. The lighting was maintained at 50-70 foot-candles on a 16-hour daylight photoperiod. An initial range-finding test was conducted to determine the concentration range for the definitive study. The preliminary test concentrations were set at 0.1, 1.0, and 10 mg/L. Based on the results of the preliminary testing, 5 test concentrations were selected and tested in duplicate, 0 (control), 0.56, 1.0, 1.8, 3.2, 5.6. and 10 mg/L.

Test concentrations were prepared by preparing a stock solution in deionized water and serially diluting to obtain desired concentrations. All results were based on the nominal concentrations. Water quality parameters of temperature, dissolved oxygen, and pH were measured at the termination of the test and were within acceptable limits.

Statistical analysis of the concentration versus effect data was obtained by employing a computerized LC50 program developed by Stephan. This program calculated the LC50 statistic and its 95% confidence limits using the binomial and moving average tests. The method of calculation selected for use was that which gave the narrowest confidence limits for the LC50.

Result

Water quality parameters of temperature, dissolved oxygen, and pH were measured at the termination of the test and were within acceptable limits. The dissolved oxygen concentrations, which ranged between 8.4 and 8.8 mg/L, were considered adequate for testing. The pH values of the treated chambers were consistent with the control and ranged from 8.2 to 8.7. The no-effect concentration based on the lack of mortality and abnormal effects was 3.2 mg/L. The abnormal effects of mortality and/or daphnids lying on the bottom were observed after 24 and 48 hours of exposure in the 5.6 mg/L (24-hour: 2/20 dead; 48-hour: 3/20 dead) and 10 mg/L (24-hour:

Test substance Reliability

15/20 dead; 48-hour: 20/20 dead) test concentrations. 2-Amino-2,3-dimethylbutanenitrile, purity not reported

(2) valid with restrictions Klimisch code: 2c

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11.05.2006 (6)

Type

Species

Daphnia sp. (Crustacea)

Exposure period

48 hour(s)

Unit **EC50** mg/l

Method

10.4

other: ECOSAR v0.993

Year **GLP**

Test substance

as prescribed by 1.1 - 1.4

Remark

Reliability: Estimated value based on accepted model.

Result

10.4 mg/L; $\log \text{Kow} = 0.87$

11.05.2006

(26)

4.3 TOXICITY TO AQUATIC PLANTS E.G. ALGAE

Species

Selenastrum capricornutum (Algae)

Endpoint

Exposure period

96 hour(s) mg/l

Unit **EC50**

.36

Method

other: Patterned after EPA 600/9-78-016/OTS/ASTM. ABC Laboratories

Protocol 8004.

Year

1984

GLP Test substance ves as prescribed by 1.1 - 1.4

Method

Patterned after EPA 600/9-78-016/OTS/ASTM. ABC Laboratories Protocol

8004.

Temperature and light readings were measured throughout the test and were within acceptable limits. The static algal toxicity study was conducted in 250 mL Erlenmeyer flasks containing 100 mL of synthetic algal nutrient medium. This media was composed of 1.0 mL of a salt solution diluted to a final volume of 1000 mL of deionized water. The deionized water was filtered through a Millipore Milli-Q water purification system. After the media was prepared, the pH was adjusted to 7.5 and filter-sterilized through a 0.45 um filter. To each flask was added 1 mL of algal inoculum containing 2x10E6±10% cells. The test vessels were incubated for 96 hours at 24±2°C under continuous "cool white" fluorescent light (maintained at 400±10% ft-c) and constant shaking. Temperature and light intensity were monitored throughout the study. Log phase growth was confirmed at 96-hours with a count of 6.9x10E5 cells/mL in the control. A 96-hour range finding study was conducted to determine the concentration range for the definitive study. Based on the results of the range finder, test concentrations were set at 0, 0.01, 0.1, 0.5, 1.0, and 10 mg/L, and test concentrations were corrected for 94.2% purity. Test flasks were prepared in triplicate for each test concentration and the control. Test concentrations were prepared by preparing a stock solution in deionized water and serially diluting to obtain desired concentrations. Statistical analysis of the concentration versus effect data was obtained by employing a computerized EC50 program developed by Stephan, performing the binomial, moving average, and probit tests. This program calculated the EC50 statistic and its 95% confidence limits using the moving average test. The method of calculation selected for use was that which gave the narrowest confidence limits for the EC50. The no effect level was determined by using ANOVA and a multiple means comparison test

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(Fisher's LSD).

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Result 0.36 mg/L (confidence limits, 0.24-0.52 mg/L) Gravimetric determinations of algal growth at each test concentration (0, 0.01, 0.1, 0.5, 1.0, and 10 mg/L) indicated percent effected as 7, 7, 7, 58, 95, and 100, respective to the concentrations tested. The no-effect level for the test compound was 0.10 mg/L. Test substance : 2-Amino-2,3-dimethylbutanenitrile, purity 94.2% : (2) valid with restrictions Reliability Klimisch code: 2c 31.05.2006 (7) **Species** other algae **Endpoint** Exposure period : 96 hour(s) Unit : mg/l **EC50** 13.3 Method : other: Modeled by ECOSAR v.0.993 Year **GLP** : no Test substance : as prescribed by 1.1 - 1.4 Remark Reliability:Estimated value based on accepted model. 13.3 mg/L; $\log Kow = 0.87$ Result 17.02.2006 (26)4.4 TOXICITY TO MICROORGANISMS E.G. BACTERIA 4.5.1 CHRONIC TOXICITY TO FISH 4.5.2 CHRONIC TOXICITY TO AQUATIC INVERTEBRATES 4.6.1 TOXICITY TO SEDIMENT DWELLING ORGANISMS 4.6.2 TOXICITY TO TERRESTRIAL PLANTS 4.6.3 TOXICITY TO SOIL DWELLING ORGANISMS 4.6.4 TOX. TO OTHER NON MAMM, TERR, SPECIES 4.7 BIOLOGICAL EFFECTS MONITORING 4.8. BIOTRANSFORMATION AND KINETICS

4.9 ADDITIONAL REMARKS

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5.0 TOXICOKINETICS, METABOLISM AND DISTRIBUTION

5.1.1 ACUTE ORAL TOXICITY

Type

LD50

Value

83 mg/kg bw

Species

Strain

Sprague-Dawley

Sex

male

Number of animals

Vehicle

Doses

31.3, 62.5, 125 mg/kg

Method Year

other 1983

GLP

no

Test substance

as prescribed by 1.1 - 1.4

Method

: Animals were housed at room temperature, 5/cage, and were fasted 18 hour before dosing. Test material was suspended in corn oil. Ten male rats received neat 2-amino-2,3-dimethylbutanenitrile by gavage in corn oil (5% w/v) at concentrations of 31.3, 62.5, and 125 mg/kg. Animals were dosed by oral gavage and observed several times after dosing, and twice daily over a 14-day period for physical condition and mortality.

Result

Toxic signs seen in all 10 animals at the highest dose and in 1 animal at the intermediate dose included tremors, tonic convulsions, salivation, and prostration. All animals in the 125 mg/kg dose group and 1 of the rats in the 62.5 mg/kg dose group died within 8 hours of dosing.

Test substance

2-Amino-2,3-dimethylbutanenitrile, purity>95%

Reliability

(2) valid with restrictions

Klimisch code: 2e. This study was not conducted under GLP or OECD guidelines, but generally meets scientific standards, is well documented.

and is accepted for assessment.

11.05.2006

(4)(12)

5.1.2 ACUTE INHALATION TOXICITY

Type

LC50

Value

73 ppm

Species

rat

Strain

Sprague-Dawley

Sex

male/female

Number of animals

Vehicle

Doses

Exposure time

4 hour(s)

Method

OECD Guide-line 403 "Acute Inhalation Toxicity"

Year

1988 ves

GLP Test substance

as prescribed by 1.1 - 1.4

Method

: OECD Guideline 403 "Acute Inhalation Toxicity"

Each group, containing 5 male and 5 female rats, was exposed once for 4

hours to vapor dynamically generated from 2-amino-2,3-

dimethylbutanenitrile. The chamber atmosphere was monitored for 2amino-2,3-dimethylbutanenitrile and hydrogen cyanide. Body weight gains

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Result

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were observed for all survivors on days 7 and 14. Clinical signs and

macroscopic findings were recorded. 73 ppm (confidence limits, 67-79 ppm)

The mean concentrations of 2-amino-2,3-dimethylbutanenitirle and (HCN) for the four 4-hour exposures were 77 (6), 71 (8), 58 (4), and 21 (<2) ppm. Mortality was observed in the 71 (40%) and 77 (70%) ppm groups. All deaths occurred on the day of exposure. Clinical signs were observed on the day of exposure for all groups except the 21 ppm group and included

hypoactivity, ataxia, prostration, and signs of respiratory irritation. Hypoactivity during exposure was the only clinical sign seen in rats in the 58 ppm group. Animals were observed for the 14-day post-exposure period, and had no clinical signs of toxicity. No macroscopic lesions were observed in the remaining rats that died or in the rats killed at the end of

the 2-week recovery period.

Test substance Reliability

: 2-Amino-2,3-dimethylbutanenitrile, 96% in toluene

: (1) valid without restriction

Klimisch code: 1a. This study was conducted under OECD guidelines.

20.02.2006

Species

Type : LC50 **Value** : 92 ppm

Strain : Sprague-Dawley
Sex : male/female

Number of animals

Vehicle Doses

1 hour(s)

10

Exposure time
Method

OECD Guide-line 403 "Acute Inhalation Toxicity"

Year

Method

GLP : yes
Test substance : other TS

OECD Guideline 403 "Acute Inhalation Toxicity"

Each group, containing 5 male and 5 female rats, was exposed once for 1 hour to vapor dynamically generated from 2-amino-2,3-

dimethylbutanenitrile. The chamber atmosphere was monitored for 2-

amino-2,3-dimethylbutanenitrile and hydrogen cyanide.

Result

The mean concentrations of 2-amino-2,3-imethylbutanenitrile and (HCN) for the three 1-hour exposures were 109 (12), 75 (4), and 63 (3) ppm. Mortality was observed in the 109 ppm group (9/10 rats died). All deaths occurred on the day of exposure. Clinical signs were observed on the day

of exposure for all groups except the 63 ppm group and included

hypoactivity, ataxia, prostration, and signs of respiratory irritation. Animals were observed for the 14-day post-exposure period and had no clinical signs of toxicity. Body weight gains were observed for all survivors on days 7 and 14. No macroscopic lesions were observed in the remaining rats

that died or in the rats killed at the end of the 2-week period.

Test substance Reliability

2-Amino-2,3-dimethylbutanenitrile, 96% in toluene

: (1) valid without restriction

Klimisch code: 1a. This study conducted under OECD guidelines.

11.05.2006

(10)

(10)

5.1.3 ACUTE DERMAL TOXICITY

Type : LD50

Value : 23 mg/kg bw

Species : rabbit

Strain : New Zealand white

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Sex

male

Number of animals

Vehicle

Doses

12.5, 25, 50, 100, 200 mg/kg other

Method Year **GLP**

1983 no

Test substance

as prescribed by 1.1 - 1.4

Method

Rabbits were individually quarantined 3 days prior to the Method: test. Animals were fed ad libitum during guarantine and the study. On the day prior to test, the animals were shaved. Neat test substance was applied at doses of 12.5, 25, 50, 100, and 200 mg/kg to the shaved skin of 5 groups of 5 male albino rabbits, then covered with an occlusive wrap for 24 hours. The test site was wiped clean after a 24-hour exposure period. Animals were observed for physical condition and mortality on the day of test material application and twice daily for 14 days. Gross autopsy was

not performed.

Result

All deaths occurred within 24 hours of dose application. All of the animals in the 200, 100, and 50 mg/kg dose groups died. Three of 5 rabbits in the 25 mg/kg dose group died. Signs of toxicity observed in all animals at all

dose levels included ataxia and prostration. 23 mg/kg bw (confidence interval, 16-32 mg/kg)

Exposure Time:

24 hours

Test substance Reliability

2-Amino-2.3-dimethylbutanenitrile, purity >95%

(2) valid with restrictions

Klimisch code: 2e. This study was not conducted under GLP or OECD quidelines, but generally meets scientific standards, is well documented.

and is accepted for assessment.

11.05.2006

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11.05.2006

(9)

5.1.4 ACUTE TOXICITY, OTHER ROUTES

rat

5.2.1 SKIN IRRITATION

Species

Concentration **Exposure**

Exposure time Number of animals

Vehicle PDII

Result Classification

Method Year 1984 GLP yes

Test substance

as prescribed by 1.1 - 1.4

Method

: A 28-day repeated dermal neurotoxicity study was conducted to assess the potential of the test substance to cause systemic toxicity and adverse effects on the nervous system. The test substance was administered dermally to rats (5/sex/group) at concentrations of 0, 3, 10, and 30 mg/kg (0, 3.578, 11.932, and 35.775 uL/kg) for 6 hours/day, 5 days/week for 4

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weeks. The test substance was applied by gentle inunction over the clipped area of unabraded skin. Dosages were adjusted at 3-day intervals to accommodate body weight changes. The treated area was covered with an impervious patch. After 6 hours, the patch was removed and the

treated area thoroughly cleansed.

Remark For additional details regarding methods for the subchronic study, refer to

Section 5.4

For additional details regarding subchronic results of this study, refer to

Section 5.4

Result : All rats survived the experimental period. Skin irritation, consisting of mild

erythema, eschar formation, dry and/or flaky skin, and small sores were observed at the application site of rats in the 10 and 30 mg/kg dose groups.

No significant irritation was seen in the rats in the 3 mg/kg dose group.

Test substance

2-Amino-2,3-dimethylbutanenitrile, purity 94.2%

Reliability

(1) valid without restriction

Klimisch code: 1b. This study was not conducted under OECD guidelines,

but was conducted under GLP.

15.05.2006

5.2.2 EYE IRRITATION

Species Concentration rabbit 89 mg

Dose

Exposure time

Comment

Number of animals

Vehicle Result

Classification

Method

Year **GLP**

other

1988 no data

Test substance

as prescribed by 1.1 - 1.4

Method

The test substance (89 mg) was instilled into the eyes of rabbits. No

additional details were reported.

Remark

Reliability: Not assignable because insufficient study information was

Result

Instillation of 89 mg of the test substance into the eyes of rabbits resulted in

the death of 5 of 6 rabbits tested. No additional data was reported.

Test substance

: 2-Amino-2-methylbutanenitrile, purity >95%

11.05.2006

(12)

5.3 SENSITIZATION

5.4 REPEATED DOSE TOXICITY

Type

Species

rat

:

Sex

: male/female

Strain

other: Rats/Charles River CD (Sprague-Dawley derived)

Route of admin. Exposure period dermal 28 days

Frequency of treatm.

6 hours/day, 5 days/week

Post exposure period Doses

0, 3, 10, 30 mg/kg

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Control group

yes

NOAEL Method

3 mg/kg bw other

Year **GLP**

1984 yes

Test substance

as prescribed by 1.1 - 1.4

Method

A 28-day repeated dermal neurotoxicity study was conducted to assess the potential of the test substance to cause systemic toxicity and adverse effects on the nervous system. The test substance was administered dermally to rats (5/sex/group) at concentrations of 0, 3, 10, and 30 mg/kg (0, 3.578, 11.932, and 35.775 uL/kg) for 6 hours/day, 5 days/week for 4 weeks. The test substance was applied by gentle inunction over the clipped area of unabraded skin. Dosages were adjusted at 3-day intervals to accommodate body weight changes. The treated area was covered with an impervious patch. After 6 hours, the patch was removed and the treated area thoroughly cleansed. Detailed observations, body weights, and food consumption values were recorded at 3-day intervals.

Animals were perfused with 10% buffered neutral formalin solution prior to necropsy. The weights of the liver, kidney, heart, thyroid glands, brain, and gonads were recorded.

Result

All rats survived the experimental period. There were no overt signs of toxicity observed at any treatment level; body weight gain, diet consumption, hematology, and clinical chemistry values were comparable across all groups.

A statistically significant increase in absolute thyroid weights was observed in male rats at all treatment levels. Thyroid weights for females were somewhat increased, though not significantly. Relative thyroid weights were also somewhat increased at all levels in both sexes with a significant increase in males at the 3 mg/kg level. Subsequent histopathology failed to find any pathologic change that would account for this finding. No other significant organ weight changes were observed at any treatment level. No test article-related gross or microscopic lesions were observed in the tissue samples from the adrenal gland, bone marrow, brain, eye and optic nerve, heart, liver, kidneys, lung, ovary, skeletal muscle, sciatic nerve, skin, spinal cord, testes, thyroid glands, and uterus. There were no overt signs of neurotoxicity at any treatment level. Skin irritation, consisting of mild erythema, eschar formation, dry and/or flaky skin, and small sores were observed at the application site of rats in the 10 and 30 mg/kg dose groups. No significant irritation was seen in the rats in the 3 mg/kg dose group.

The authors of this study, therefore, concluded that the NOEL is 3 mg/kg. As the intent of the repeated exposure dermal study is to assess systemic toxicity following dermal application of the test material, and as no evidence of systemic toxicity was observed at the high dose, one could conclude that 30 mg/kg did not produce systemic toxicity or neurotoxicity and should be considered a NOEL.

Test substance Reliability

2-Amino-2,3-dimethylbutanenitrile, purity 94.2%

(1) valid without restriction

Klimisch code: 1b. This study was not conducted under OECD guidelines, but was conducted under GLP.

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(8)

Type **Species** Sex

Sub-chronic

Strain

other: Charles River CD rats

Route of admin. **Exposure period** Frequency of treatm.

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Post exposure period

Doses

Control group

Method

Year **GLP**

yes

Test substance

other TS

Method

: A 28-day repeated dermal neurotoxicity study was conducted to assess the potential of the test substance to cause systemic toxicity and adverse effects on the nervous system. The test substance was administered dermally to rats (5/sex/group) at concentrations of 0, 3, 10, and 30 mg/kg (0, 3.578, 11.932, and 35.775 ?L/kg) for 6 hours/day, 5 days/week for 4 weeks. The test substance was applied by gentle inunction over the clipped area of unabraded skin. Dosages were adjusted at 3-day intervals to accommodate body weight changes. The treated area was covered with an impervious patch. After 6 hours, the patch was removed and the

treated area thoroughly cleansed.

Result

All rats survived the experimental period. Skin irritation, consisting of mild erythema, eschar formation, dry and/or flaky skin, and small sores were observed at the application site of rats in the 10 and 30 mg/kg dose groups. No significant irritation was seen in the rats in the 3 mg/kg dose group.

Test substance Reliability

2-Amino-2,3-dimethylbutanenitrile, purity 94.2%

(1) valid without restriction

Klimisch code: 1b. This study was not conducted under OECD guidelines.

but was conducted under GLP.

11.05.2006

(8)

5.5 GENETIC TOXICITY 'IN VITRO'

Bacterial reverse mutation assay

System of testing Test concentration

Salmonella typhimurium TA98, TA100, TA1535, TA1537 0.1, 1, 10, 100 ug/plate (0.1 uL test substance/plate)

Cycotoxic concentr. Metabolic activation

with and without negative

Result Method

EPA OPPTS 870.5265

Year **GLP**

1983

Test substance

as prescribed by 1.1 - 1.4

Method

EPA OPPTS 870.5265

The maximum concentration tested in the Ames Salmonella Plate assay with and without metabolic activation (S-9) using bacterial strains TA98. TA100, TA1535, and TA1537 was 5000 ug/plate. The positive controls were 2 aminoanthracene (2-AA), N-methyl-N-nitro-N-nitrosoguanidine (MNNG), 9-aminoacridine (9-AA), and 2 nitrofluorene (2-NF). The negative

(solvent) control was ethanol. Remark

The test substance was cytotoxic at 1000 and 5000 ug/plate. No evidence

of base-pair substitution or frame-shift mutation was observed.

Test substance

2-Amino-2,3-dimethylbutanenitrile, purity not reported

(2) valid with restrictions Reliability

Klimisch code: 2e. This study was not conducted under GLP or OECD guidelines, but generally meets scientific standards, is well documented.

and is accepted for assessment.

11.05.2006

(2)

11.05.2006

(12)

5.6 GENETIC TOXICITY 'IN VIVO'				(1) (1) (1) (1) (1) (1) (1) (1) (1) (1)	
5.7 CARCINOGENICITY					
5.8.1 TOXICITY TO FERTILITY					
5.8.2 DEVELOPMENTAL TOXICITY/TERA	TOGENICI	Ŋ			
5.8.3 TOXICITY TO REPRODUCTION, OTH	IER STUD	IES			
5.9 SPECIFIC INVESTIGATIONS			e de la companya de		
5.10 EXPOSURE EXPERIENCE					
5.11 ADDITIONAL REMARKS					7

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5. Toxicity

6. Analyt. Meth. for Detection and Identification

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6.1 ANALYTICAL METHODS

6.2 DETECTION AND IDENTIFICATION

	7. Eff.	Agains	t Target	Org. and	Intended	i Uses	13893-53-3 02.06.2006
7.4 USER RESISTANCE	7.1 I	FUNCTION					
7.5 RESISTANCE	7.2 i	EFFECTS (ON ORGANI	ISMS TO BE	CONTROLLE	D	
7.5 RESISTANCE	73 (ORGANISM	ISTO BE P	ROTECTED			
	74 (USER					
	75 I	RESISTAN	Ó E				
						÷	
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	,						

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8. Meas. Nec. to Prot. Man, Animals, Environment		13893-53-3 02.06.2006
8.1 METHODS HANDLING AND STORING		
8.2 FIRE GUIDANCE		學的學生
8.3 EMERGENCY MEASURES		
8.4 POSSIB. OF RENDERING SUBST. HARMLESS		
8.5 WASTE MANAGEMENT		
8.6 SIDE-EFFECTS DETECTION 4.		
8.7 SUBSTANCE REGISTERED AS DANGEROUS FOR GROUND WAT	IER 🗇	
8.8 REACTIVITY TOWARDS CONTAINER MATERIAL		

9. References Id 13893-53-3
Date 02.06.2006

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- 10.1 END POINT SUMMARY
- 10.2 HAZARD SUMMARY
- 10.3 RISK ASSESSMENT